

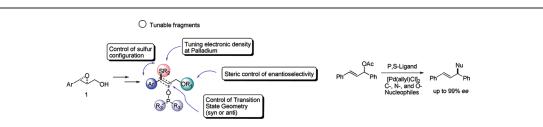
Phosphinite Thioethers Derived from Chiral Epoxides. Modular *P*,*S*-Ligands for Pd-Catalyzed Asymmetric Allylic Substitutions

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A new family of modular *P*,*S*-ligands has been prepared from enantiopure arylglycidols. These ligands have been iteratively optimized with respect to four different structural parameters for use in Pd-catalyzed allylic substitutions. As a final output, highly active and enantioselective ligands for these synthetically important transformations have been developed, and the factors controlling their catalytic behavior have been rationalized. From a methodological point of view, a convenient procedure for the regioselective ring-opening of *cis*-glycidic esters with bulky thiols to yield the corresponding β -alkylthio- α -hydroxy carboxylic acids has been developed.

Introduction

Metal-catalyzed asymmetric reactions have become one of the most powerful tools for the production of enantiomerically enriched compounds. Although the use of numerous chiral ligands has been reported, the design and synthesis of new types of ligands with improved performance continues to attract the interest of synthetic chemists.¹

In general, the design of new catalysts makes use of either C_2 -symmetrical ligands, resulting in catalytic systems with restricted numbers of competing diastereomeric transition states,² or involves ligands containing different donor atoms, able to generate electronic asymmetry on the metal. This asymmetry can be transmitted to reacting molecules bonded to the metal (for instance, through the *trans* effect)³ and has the potential to control both the stability and reactivity of metal–substrate intermediates.

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While the use of bidentate ligands with P,N coordination mode has become a mature tool in asymmetric catalysis,⁴ the use of related P,S-ligands has gained considerable momentum in recent times.⁵ These ligands present a key structural property that makes them very attractive in asymmetric

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 ^{(1) (}a) Comprehensive Asymmetric Catalysis; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Berlin, 1999; Vols. I–III. (b) Asymmetric Catalysis in Organic Synthesis; Noyori, R., Ed.; Wiley: New York, 1994.
 (c) Asymmetric Synthesis, 2nd ed.; Ojima, I., Ed.; VCH: New York, 2000.
 (d) Catalytic Asymmetric Synthesis. Acc. Chem. Res. 2000, 33(6) (special issue).

⁽²⁾ Whitesell, J. K. *Chem. Rev.* **1989**, *89*, 1581.
(3) (a) Appleton, T. G.; Clark, H. C.; Manzer, L. E. *Coord. Chem. Rev.* **1973**, *10*, 335. (b) Murray, S.; Hartley, F. *Chem. Rev.* **1981**, *81*, 365.

^{(4) (}a) Lautens, M. A.; Pfaltz, A. Allylic substitution reactions. In *Comprehensive Asymmetric Catalysis*; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Heidelberg, 1999; Vol. 2, Chapter 24. (b) For a review of chiral P,N-ligands, see: Guiry, P. J.; Saunders, C. P. *Adv. Synth. Catal.* **2004**, *346*, 497.

⁽⁵⁾ For some recent publications of chiral P,S-ligands for asymmetric catalysis, see: (a) Herrman, J.; Pregosin, P. S.; Salzmann, R. Organometallics **1995**, 14, 3311. (b) Hiroi, K.; Suzuki, Y. Tetrahedron Lett. **1998**, 39, 6499. (c) Enders, D.; Peters, R.; Runsink, J.; Bats, J. W. Org. Lett. **1999**, 11, 1863. (d) Hauptman, E.; Fagan, P. J.; Marshall, W. Organometallics **1999**, 18, 2061. (e) Pàmies, O.; Diéguez, M.; Net, G.; Ruiz, A.; Claver, C. Organometallics **2000**, 19, 1488. (f) Evans, D. A.; Campos, K. R.; Tedrow, J. S.; Michael, F. E.; Michel, R. G. J. Am. Chem. Soc. **2000**, 122, 7905. (g) Nakano, H.; Okuyama, Y.; Hongo, H. Tetrahedron Lett. **2000**, 41, 4615. (h) Pàmies, O.; Van Strijdonck, G. P.; Diéguez, M.; Deerenberg, S.; Net, G.; Ruiz, A.; Claver, C.; Kamer, P. C.; van Leeuwen, P. W. N. M. J. Org. Chem. **2001**, 66, 8867. (i) Alexakis, A.; Benhaim, C. Tetrahedron: Asymmetry **2001**, 12, 1151. (j) García Mancheño, O.; Priego, J.; Cabrera, S.; Gómez Arrayás, R.; Llamas, T.; Carretero, J. C. J. Org. Chem. **2003**, 68, 3679. (k) Nakano, H.; Yokoyama, J.; Okuyama, Y.; Fujita, R.; Hongo, H. Tetrahedron: Asymmetry **2003**, 14, 2361. (l) Tu, T.; Zhou, Y.; Hou, X.; Xin, L.; Dong, X.; Yu, Y.; Sun, J. Organometallics **2003**, 22, 1255. (m) Evans, D. A.; Michael, F. E.; M. A.; Riera, A. J. Org. Chem. **2004**, 69, 8053. (o) García Mancheño, O.; Gómez Arrayás, R.; Carretero, J. C. J. Am. Chem. Soc. **2004**, 126, 456. (p) Molander, G. A.; Burke, J. P.; Carroll, P. J. J. Org. Chem. **2004**, 69, 8053. (o) García Mancheño, O.; Gómez Arrayás, R.; Carretero, J. C. J. Am. Chem. Soc. **2004**, 126, 456. (p) Molander, G. A.; Burke, J. P.; Carroll, P. J. J. Org. Chem. **2004**, 69, 8063. (o) García Mancheño, O.; Gómez Arrayás, R.; Carretero, J. C. J. Am. Chem. Soc. **2004**, 126, 456. (p) Molander, G. A.; Burke, J. P.; Carroll, P. J. J. Org. Chem. **2004**, 69, 8057. (o) García Mancheño, O.; Gómez Arrayás, R.; Carretero, J. C. J. Am. Chem. Soc. **2004**, 126, 456. (p) Molander, G. A.; Burke, J. P.; Carroll, P. J. J. Org. Chem. **2004**

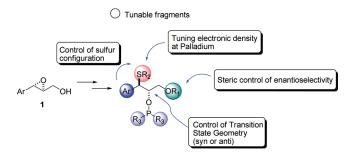


FIGURE 1. Highly modular *P*,*S*-ligands from epoxy alcohols 1.

catalysis: the sulfur atom in dissymmetrically substituted thioethers becomes stereogenic upon coordination to the metal, thus imposing a unique asymmetric environment close to the reactive metal center.⁶

In the past several years, we have successfully developed and fine-tuned families of modular bidentate ligands $(N/O, ^7 N/N, ^8 S/N, ^9 P/N, ^{10} P/O-P^{11})$ for a variety of enantioselective processes starting from purely synthetic yet enantiomerically pure precursors. In the present paper, we report the synthesis of highly modular *P*,*S*-ligands from readily available enantiopure arylglycidols (Figure 1).¹² An iterative optimization process involving up to four structural parameters has allowed the identification of ligands exhibiting high catalytic activity and enantioselectivity in the asymmetric substitution of allylic substrates with a variety of C-, N-, and O-nucleophiles.

It is to be mentioned that the use of chiral epoxides as starting materials for the modular construction of enantiopure *P*,*S*-ligands has not been reported in the literature. In a related approach, Evans et al. reported the desymmetrization¹³ of *meso*-cyclohexene oxide with *tert*-butyl thiol, followed by phosphinylation of the resulting alcohol as an entry to ligands for the allylic substitution^{5f} and olefin hydrogenation.^{5m} In the same study,^{5f} the preparation of modular phosphinito thioether ligands was performed from enantiopure *N*-acyloxazolidinones through rather complex sequences starting with a highly diastereoselective electrophilic halogenation and often involving diastereomer separation, and the effect of different structural parameters on catalytic activity and enantioselectivity was analyzed (Figure 2).

Results and Discussion

Ligand Design. The general strategy for the synthesis of these thioether–phosphinite ligands is presented in Scheme 1.

(8) Pericàs, M. A.; Puigjaner, C.; Riera, A.; Vidal-Ferran, A.; Gómez, M.; Jimenéz, F.; Muller, G.; Rocamora, M. *Chem.—Eur. J.* **2002**, *8*, 4164.

- (10) Popa, D.; Puigjaner, C.; Gómez, M.; Benet-Buchholz, J.; Vidal-Ferran, A.; Pericas, M. A. Adv. Synth. Catal. 2007, 349, 2265.
- (11) Fernández-Pérez, H.; Pericàs, M. A.; Vidal-Ferran, A. Adv. Synth. Catal. 2008, 350, 1984.

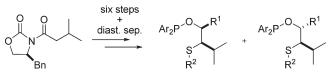
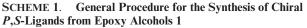
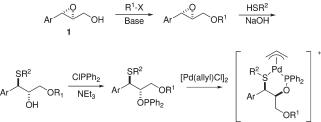
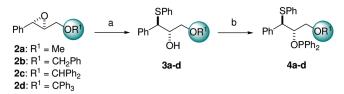


FIGURE 2. Evans' approach to modular *P*,*S*-ligands from enantiopure *N*-acyloxazolidinones.





SCHEME 2. Two-Step Synthesis of Thioether–Phosphinite Ligands 4 from Epoxy Ethers 2^{*a*}



^aReagents and conditions: (a) PhSH (2 equiv), NaOH (2 equiv), 65 °C, 20–120 min in dioxane/water (10:1); (b) ClPPh₂ (1.1 equiv), NEt₃ (1.2 equiv), 10 mol % of DMAP, 20 min in toluene, rt.

In the first step, the primary hydroxy group in the starting epoxy alcohols (1) is protected, and the resulting epoxy ethers are then subjected to regio- and stereoespecific ring-opening by thiolates to afford β -hydroxy sulfides. Treatment of these intermediates with chlorodiphenylphosphine and triethylamine in the presence of a 10% molar amount of 4-DMAP affords the desired ligands. Since diphenylphosphinite units show optimal performance in Pd-catalyzed allylic alkylation, ^{5f,q} this module was excluded from the optimization process.

For the subsequent evaluation in catalysis, the corresponding π -allylpalladium complexes were generated in situ from the corresponding *P*,*S*-ligand and metal precursor.

Ligand Optimization for the Palladium-Catalyzed Allylic Alkylation. Optimization of the Ether Moiety. We first investigated the effect of the steric bulk of the ether moiety OR^1 on the catalytic properties (activity and enantioselectivity) of the derived π -allylpalladium complexes. The steric environment at this position has proved to exert a critical influence on the catalytic activity of other chiral ligands derived from Sharpless epoxy alcohols in a variety of asymmetric reactions.^{7,8,10,11}

The starting point in our optimization process was *trans*phenylglycidol (**1a**; Ar = Ph), which is readily available in enantiopure form by Sharpless epoxidation and can be modified to incorporate a variety of OR¹ groups (Scheme 2). According to our synthetic plan, epoxy ethers bearing alcohol protecting groups of increasing sizes (**2a**-**d**) were synthesized,^{7a} and then the regioselective and stereospecific ring-opening of these epoxy ethers with benzenethiol was studied.

⁽⁶⁾ For a reviews of chiral sulfur ligands in asymmetric catalysis, see:
(a) Mellah, M.; Voituriez, A.; Schulz, E. Chem. Rev. 2007, 107, 5133.
(b) Pellissier, H. Tetrahedron 2007, 63, 1297. (c) Bayon, J. C.; Claver, C.; Masdeu-Bultó, A. Coord. Chem. Rev. 1999, 193-195, 73. (d) Masdeu-Bultó, A.; Diguez, M.; Martin, E.; Gómez, M. Coord. Chem. Rev. 2003, 242, 159.

 ^{(7) (}a) Vidal-Ferran, A.; Moyano, A.; Pericàs, M. A.; Riera, A. J. Org. Chem. 1997, 62, 4970. (b) Pastó, M.; Riera, A.; Pericàs, M. A. Eur. J. Org. Chem. 2002, 2237. (c) Puigjaner, C.; Vidal-Ferran, A.; Moyano, A.; Pericàs, M. A.; Riera, A. J. Org. Chem. 1999, 64, 7902.

⁽⁹⁾ Jimeno, C.; Moyano, A.; Pericàs, M. A.; Riera, A. Synlett 2001, 7, 1155.

⁽¹²⁾ Katsuki, T.; Martin, V. S. Org. React. 1996, 48, 1-299.

⁽¹³⁾ Iida, T.; Yamamoto, N.; Sasai, H.; Shibasaki, M. J. Am. Chem. Soc. 1997, 119, 4783.

 TABLE 1.
 Yields for the Synthesis of Thioether-Phosphinite Ligands 4a-d

entry	epoxy ether	yield 3 (%)	yield 4 (%)	overall yield (%)
1	2a	90 ^a	91	82
2	2b	72^{b}	82	59
3	2c	90^c	85	77
4	2d	74^d	88	65
^a 20 1	min. ^{<i>b</i>} 60 min. ^{<i>c</i>} 12	20 min. ^{<i>d</i>} 120 mir	n, PhSH (5 equiv	v), NaOH (5 equiv).

One of the most straightforward synthetic procedures for the ring-opening of epoxides with thiols takes place under basic conditions in a protic solvent.¹⁴ Alternatively, several methods have been reported involving Lewis acid catalysis, such as the use of catalytic amounts of $InCl_3^{15a}$ or metal perchlorate salts in combination with thiols.^{15b} Although the use of $InCl_3$ or perchlorates has been reported to induce a completely regioselective ring-opening at the benzylic position, a mixture of diastereomers was obtained in our case, probably via participation of an S_N1 mechanism.¹⁶ Very gratifyingly, when phenylglycidyl ethers **2a**–**d** were treated with sodium benzenethiolate in dioxane–water,^{14a} a completely regioselective and stereospecific ring-opening took place. A final phosphinylation led to thioether phosphinites **4a**–**d**, as shown in Scheme 2.

It is interesting to note that the steric effect of \mathbb{R}^1 has some influence on the reactivity of the epoxy ethers. For instance, while the reaction of **2a** ($\mathbb{R}^1 = Me$) proceeded smoothly (90% yield) in 20 min (entry 1), **2d** ($\mathbb{R}^1 = CPh_3$) required a larger excess of thiol (5 equiv) and longer reaction time (2 h) to achieve a satisfactory conversion (entry 4). The obtained β -hydroxy sulfides were derivatized with chlorodiphenylphospine in the presence of triethylamine and DMAP following a well-established procedure for related compounds.¹⁷ The target ligands **4a**–**d** were obtained with overall yields ranging from 59% to 82% (Table 1) after purification by filtration through a short pad of silica gel.

Ligands $4\mathbf{a} - \mathbf{d}$ were next evaluated in the palladiumcatalyzed asymmetric allylic alkylation (AAA) of 1,3-diphenylpropenyl acetate (S1) with dimethyl malonate, using *N*,*O*-bis(trimethylsilyl)acetamide (BSA) as a base.^{5f} The reactions were carried out in dichloromethane at room temperature in the presence of a catalyst generated in situ from 1.25 mol % of π -allylpalladium chloride dimer, 2.5 mol % of ligand, and a catalytic amount of KOAc (Table 2).

As clearly shown in Table 2, complete conversions were observed after 20 min at room temperature for ligands 4c and 4d and almost complete conversion for 4a and 4b. Unfortunately, ligands 4a-d gave the alkylation product with poor enantioselectivities (10-12% ee) (Table 2).

Although these results were very disappointing in terms of enantioselectivity, they provided initial indication that the steric bulk of R^1 has no effect on the stereochemical outcome of the reaction. Thus, this fragment could be fixed

TABLE 2.	Effect of the Steric Bulk in R ¹ on Conversion and
Enantioselec	ivity

OAc Ph Ph	CH ₂ (CO	Ph No [Pd(allyl)Cl] ₂ OMe) ₂ (3 equiv) equiv), KOAc	CH(COOMe) ₂ Ph P1
ligand	R^1	$\operatorname{conv}^{a}(\%)$	ee^{b} (%)
4a	Me	92	12
4b	CH ₂ Ph	87	11
4c	CHPh ₂	> 99	10
4d	CPh ₃	> 99	12
^a Conversion a	fter 20 min by	¹ HNMR of the reaction	crude. ^b ee values

"Conversion after 20 min by 'H NMR of the reaction crude." ee values by chiral HPLC.

 TABLE 3.
 Effect of the Sulfur Substituent on Conversion and Enantioselectivity

OAc Ph Ph S1		Ph 2.5 mol % OPPh ₂ 1.25 mol % [Pd(allyl)Cl] ₂ CH ₂ (COOMe) ₂ (3 equiv) BSA (3 equiv), KOAc CH ₂ Cl ₂ , rt	CH(COOMe) ₂ Ph Ph Ph	
entry	ligand	\mathbb{R}^2	$\operatorname{conv}^{a}(\%)$	ee^{b} (%)
1	4e	3,5-dimethylphenyl	93	20
2	4 f	2,6-dimethylphenyl	>99	43
2 3 4 5	4g	<i>p</i> -methoxyphenyl	94	9
4	4h	<i>p</i> -bromophenyl	>99	10
5	4i	<i>p-tert</i> -butylphenyl	>99	15
6	4j	2-naphthyl	88	15
7	4k	isopropyl	>99	66
8	41	cyclohexyl	92	71
9	4m	<i>tert</i> -butyl	>99	71
10	4n	adamantyl	>99	83
		r 20 min by ¹ H NMR of the	reaction crude.	^b ee values
by chira	al HPLC.			

at convenience for the optimization of the remaining modules.

Optimization of the Sulfur Substituent. A methyl substituent was selected as \mathbb{R}^1 due to the favorable reactivity of the corresponding epoxy ether toward thiolate ring-opening. With this restriction, a variety of *S*-aryl derivatives with different substitution patterns on the aryl moiety ($4\mathbf{e}-\mathbf{j}$) and *S*-alkyl derivatives ($4\mathbf{k}-\mathbf{n}$) could be easily synthesized following the general synthetic strategy described above (Scheme 1). Under the standard conditions for the AAA, we studied the effect of the thioether substituent on the catalytic performance of ligands $4\mathbf{e}-\mathbf{n}$.

As shown in Table 3, high conversions are observed with all the studied ligands after 20 min reaction, thus indicating that the nature of the sulfur substituent does not play any role on catalytic activity. Concerning enantioselectivity, changes in the *S*-substituent showed a remarkably effect, with bulkier substituents leading to higher enantioselectivities. Thus, among the studied *S*-aryl substituents, the most notable improvement in enantioselectivity was observed

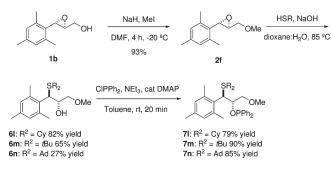
 ^{(14) (}a) Takano, S.; Yanase, M.; Ogasawara, K. *Heterocycles* 1989, 29, 249.
 (b) Corey, E. J.; Albright, J. O.; Barton, A. E.; Hashimoto, S. J. Am. Chem. Soc. 1980, 102, 1436.

^{(15) (}a) Yadav, J. S.; Reddy, B. V. S.; Baishya, G. Chem. Lett. 2002, 31, 906. (b) Chini, M.; Crotti, P.; Giovani, E.; Macchia, F.; Pineschi, M. Synlett 1992, 4, 303.

⁽¹⁶⁾ Parker, R. E.; Isaacs, N. S. Chem. Rev. 1959, 59, 737.

^{(17) (}a) Ohe, K.; Morioka, K.; Yonehara, K.; Uemura, S. *Tetrahedron: Asymmetry* **2002**, *13*, 2155. (b) Jia, X.; Li, X.; Lam, W. S.; Kok, S. H. L.; Xu, L.; Lu, G.; Yeung, C.-H.; Chan, A. S. C. *Tetrahedron: Asymmetry* **2004**, *15*, 2273.

SCHEME 3. Preparation of Ligands Bearing a Mesityl Substituent



with the bulky 2,6-dimethylphenylthio group (**4f**, entry 2). These results provided an indication that bulkier alkyl thioether substituents could lead to a further increase in enantioselectivity. In fact, secondary and tertiary *S*-alkyl substituents showed a higher enantioselectivity compared to *S*-aryl ligands. A much significant increase in the asymmetric induction was observed in the case of sterically more demanding groups such as adamantyl and *tert*-butyl, which afforded the alkylation product in 83% and 71% ee, respectively (entries 9 and 10).

As already mentioned, the sulfur atom becomes stereogenic upon coordination to the metal, although control of this new chiral center is not always attainable due to its low inversion barrier (10-15 kcal/mol).^{6d,18} In catalytic applications, this interconversion could have a negative influence on enantioselectivity due to the presence of competing diastereomeric complexes. We reasoned that sulfur configuration could be efficiently controlled when a bulky substituent on sulfur and a bulky aryl substituent are simultaneously present on the ligand molecule.^{5f} In line with this notion, we synthesized analogues of the best ligands prepared so far but bearing a more sterically demanding mesityl substituent on the hydrocarbon chain.

Optimization of the Skeletal Aryl Substituent. Following the general synthetic strategy, ligands **7**l, **7m**, and **7n** were synthesized from the chiral epoxide **1b** bearing a mesityl substituent (Scheme 3).¹⁹

Reaction of the sodium alkoxide of **1b** with methyl iodide allowed the isolation of the expected methyl ether **2f** in 93% yield. β -Hydroxy sulfides **6l**-**m** were obtained by ring-opening with thiolates,¹⁴ whereas the β -hydroxy sulfides **4l**-**n** (Ar = Ph) were obtained in good yields at 65–80 °C in 1 h; higher temperatures and longer reaction times (4 h for **6l** and 20 h for **6m** and **6n**) were required to achieve the ring-opening of epoxy ether **2f** (Ar = mesityl). Finally, the target ligands **7l**-**n** were readily prepared from β -hydroxy thioethers **6l**-**n** under the standard phosphinylation conditions (Scheme 3). Ligands **7l**-**n** were next evaluated in the AAA of **S1** (Table 4).

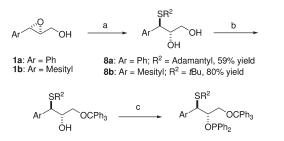
The incorporation of the bulkier mesityl substituent on the ligand structure provoked a small decrease in the reaction rate. Thus, nearly complete conversions were achieved in 45 min at room temperature for ligands 7m and 7n (entries 2, 3) and in 90 min for ligand 7l (entry 1). For comparison, the corresponding ligands (4l-n) with Ar = phenyl afforded

 TABLE 4.
 Effect of the Aryl Substituent on Conversion and Enantioselectivity

OAc Ph Ph		2.5 mol %	le → Ph	CH(COOMe) ₂	
	S1	1.25 mol % [Pd(allyl)Ci CH ₂ (COOMe) ₂ (3 equiv BSA (3 equiv), KOAc CH ₂ Cl ₂ , rt		P1	
entry	ligand	R^2	$\operatorname{conv}^{a}(\%)$	ee^{b} (%)	
1	71	cyclohexyl	96	82	
2	7m	tert-butyl	97	87	
3	7n	adamantyl	> 99	80	

^{*a*}Conversion after 90 min for ligand 71 and 45 min for ligands 7**m**,**n** by ¹H NMR of the reaction crude. ^{*b*}ee values by chiral HPLC.

SCHEME 4. Synthesis of Ligands 10a and 10b^a



^aReagents and conditions: (a) **1a**: AdSH (2 equiv), NaOH (2 equiv), dioxane/water, 80 °C, 120 min; **2a**: *t*-BuSH (3 equiv), NaOH (3 equiv), dioxane/water, 80 °C, 3 h; (b) ClCPh₃ (1.2 equiv), pyridine, 90 °C, 18 h; (c) ClPPh₂ (1.1 equiv), NEt₃ (1.2 equiv), 10 mol % 4-DMAP, toluene, rt, 40 min.

complete conversion in 20 min (Table 3). Interestingly, a positive effect on the enantioselectivity was observed for ligands 71 and 7m, affording alkylated product P1 in 82% and 87% ee (entries 1, 2). On the other hand, with ligand 7n the enantioselectivity decreased slightly (entry 3) compared with the parent ligand 4n (Ar = Ph) (Table 3, entry 10).

In view of these results, we decided to re-evaluate the effect of bulkiness on the alkoxy group (OR¹) on the best ligands prepared so far (**4n** and **7m**) in order to seek further improvement in enantioselectivity. To this end, we planned to prepare ligands **10a** and **10b**, conceptually derived from **4n** and **7m** and containing a much more sterically demanding trityloxy substituent ($\mathbf{R}^1 = \mathbf{CPh}_3$).

Unfortunately, the general synthetic strategy employed so far failed for the preparation of ligands **10a** and **10b** because ring-opening of the trityl ethers of **1a** and **1b** by the required bulky thiolates could not be accomplished. Alternatively, ligands **10a** and **10b** were prepared by a three-step sequence starting with the thiolate ring-opening of the corresponding epoxyalcohols (**1a**, **1b**). The so-prepared 3-alkylthio 1,2-diols **8a** and **8b**, were then submitted to selective protection of the primary hydroxy group with trityl chloride in pyridine at 90 °C and subsequent phosphinylation of the secondary alcohol (Scheme 4).

⁽¹⁸⁾ Abel, E.; Dormer, J.; Ellis, D.; Orrell, K. G.; Sik, V.; Hursthouse, M. B.; Mazid, M. A. J. Chem. Soc., Dalton Trans. **1992**, 1073.

⁽¹⁹⁾ Medina, E.; Moyano, A.; Pericàs, M. A.; Riera, A. Helv. Chim. Acta 2000, 80, 972.

SCHEME 5. Results obtained in the Pd-catalyzed AAA of S1 with ligands 10a and $10b^{\alpha}$



^{*a*}Reaction conditions: 1.25 mol % of $[Pd(\eta^3-C_3H_5)Cl]_2$, 2.5 mol % of **10a,b**, KOAc, dimethyl malonate/BSA (3.0 equiv), CH₂Cl₂, rt.

When **10a** and **10b** were tested in the Pd-catalyzed AAA, we were pleased to find that this modification in the \mathbb{R}^1 group had a positive influence on both catalytic activity and enantio-selectivity, with **10b** leading to the highest enantioselectivity recorded over the whole optimization process (Scheme 5).

As a final element in our screening, we decided to modify the relative stereochemistry of the C2/C3 stereogenic centers in the ligand backbone (Figure 3). It was expected that this modification could lead to substantial changes in the conformational behavior of the 6-membered ring Pd chelate and, hence, to an increased discrimination between the enantiotopic faces of the π -allyl system undergoing the substitution reaction.



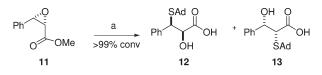
FIGURE 3. Modification of the relative stereochemistry of C2/C3.

Modification of the Relative Configuration of the Chiral Centers. The absolute and relative configurations of the two adjacent functionalized carbons present in the ring-opening products are determined by the configuration of the starting epoxide. Thus, starting from the *cis*-epoxide and following the general synthetic strategy described in Scheme 1, it should be possible to obtain the target ligands **10a**-*syn* and **10b**-*syn*; however, highly efficient and enantioselective epoxidation of *cis*-cinnamyl alcohol still remains a problem.^{20–22}

As an alternative, we planned to access the enantiopure *cis*epoxy ester through a completely stereodefined approach and to secure the target *syn* ligands through a ring-opening plus reduction sequence. For the implementation of this plan, we took advantage of the highly enantioselective catalytic dihydroxylation (ADH) of inexpensive (*E*)-methyl cinnamate. Thus, the 2,3-dihydroxy ester, obtained in high yield and with an enantiomeric excess > 99%, was converted to the *cis*glycidic ester **11** by regioselective tosylation and subsequent cyclization in basic media with an overall yield of 80%.²³

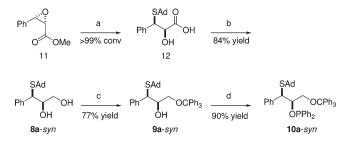
When the ring-opening of the glycidic ester **11** with adamantanethiol in the presence of sodium hydroxide was attempted, the process was poorly regioselective leading to a 2.5:1 mixture of regioisomers (Scheme 6).

SCHEME 6. Thiolate Ring-Opening of Epoxy Ester 11^a



^aReagents and conditions: (a) AdSH (3 equiv), NaOH (3 equiv), dioxane/water, 80 °C, 120 min.

SCHEME 7. Synthesis of Ligand 10a-syn^a



^aReagents and conditions: (a) AdSH (2 equiv), LiOH (2 equiv), dioxane/ water, 80 °C, 60 min; (b) $BH_3 \cdot SMe_2$ (4 equiv), THF, 16 h; (c) ClCPh₃ (1.2 equiv), pyridine, 90 °C, 18 h; (d) ClPPh₂ (1.1 equiv), NEt₃ (1.2 equiv), 10 mol % 4-DMAP, toluene, rt, 40 min.

trans-Glycidic esters and amides are known to undergo regioselective and stereoespecific opening by thiols, whereas this process is less reliable with the *cis*-isomers.²⁴ However, after some experimentation we found that the use of lithium hydroxide instead of sodium hydroxide led to ring-opening of *cis*-glycidic ester **11** with complete regioselectivity at the C3 position and in a stereoespecific manner (Scheme 7). Interestingly, crude **12** obtained by this procedure was notably clean, so that further purification was not required. Reduction of **12** with borane—dimethyl sulfide complex in THF gave **8a**-*syn* in 85% yield which was converted uneventfully into **10a**-*syn*.

In order to rationalize the regioselectivity observed in the ring-opening reaction when lithium hydroxide was used as a base, a theoretical study was carried out for the two different pathways of the reaction (attack at C2 or C3). Taking into account that ester solvolysis likely precedes ring-opening, the studied reaction system consisted in one molecule of lithium carboxylate plus one molecule of lithium thiolate. Transition states (TSs) for the epoxide openings at C2 (TS-C2) and C3 (TS-C3) were located and characterized using DFT calculations (B3LYP/6-31G (d)). In both TSs (Figure 4), one lithium cation is simultaneously coordinated to the epoxide and to one oxygen atom of the carboxylate. The other lithium cation, in turn, is coordinated to the remaining oxygen of the carboxylate and to the attacking thiolate. Very interestingly, in the transition state corresponding to the ring-opening at C3 (TS-C3) a planar arrangement of the phenyl group respect to C3 indicates the presence of carbocationic character at that position. This should be reflected in a very high reactivity toward nucleophilic attack at C3, and accordingly, a difference of $5.5 \text{ kcal} \cdot \text{mol}^{-1}$ in favor of the TS leading to the C3 product was found. If the ratio of the

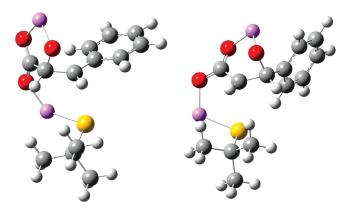
⁽²⁰⁾ Denis, J. N.; Greene, A. E.; Serra, A. A.; Luche, M.-J. J. Org. Chem. 1986, 51, 46.

⁽²¹⁾ Zhang, W.; Basak, A.; Kosugi, Y.; Hoshino, Y.; Yamamoto, H. *Angew. Chem., Int. Ed.* **2005**, *44*, 4389.

⁽²²⁾ Deng, L.; Jacobsen, E. N. J. Org. Chem. **1992**, 57, 4320.

⁽²³⁾ Wang, Z.-M.; Kolb, H. C.; Sharpless, K. B. J. Org. Chem. 1994, 59, 5104.

^{(24) (}a) Aggarwal, V. K.; Charmant, J. P. H.; Fuentes, D.; Harvey, J. N.; Hynd, G.; Ohara, D.; Picoul, W.; Robiette, R.; Smith, C.; Vasse, J. L.; Winn, C. L. J. Am. Chem. Soc. **2006**, 128, 2105. (b) Hashiyama, T.; Inoue, H.; Konda, M.; Takeda, M. J. Chem. Soc., Perkin Trans. 1 **1984**, 1725. (c) Hashiyama, T.; Inoue, H.; Takeda, M. J. Chem. Soc., Perkin Trans. 1 **1985**, 421.

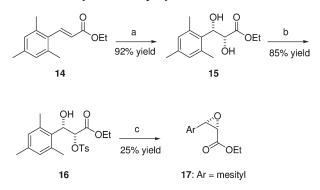


TS C3 = 0 kcal/mol

TS C2 = 5.5 kcal/mol

FIGURE 4. DFT-optimized transition states TS-C3 and TS-C2.

SCHEME 8. Synthesis of Epoxy Ester 17^a



^{*a*}Reagents and conditions: (a) see ref 25; (b) CITs (1.02 equiv), NEt₃ (1.5 equiv), CH₂Cl₂, 0 °C, 70 h; (c) K₂CO₃ (3 equiv), H₂O (5 equiv), DMF, rt, 24 h.

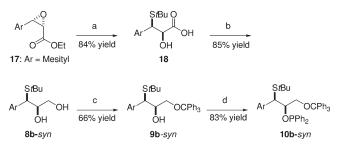
products follows a Maxwell–Boltzmann distribution based on the energies of the transition states, the theoretically computed regioisomeric ratio would be higher than 99:1, in agreement with the experimental value.

Epoxyester 17, bearing a mesityl substituent, was prepared from 14 by an analogous sequence (Scheme 8). Probably due to the presence of unfavorable steric interactions in the transition state leading to 17, the cyclization of 16 was less efficient, and the desired epoxy ester 17 was obtained in 63% yield at 40% conversion (25% yield).

Ring-opening of the rather unreactive epoxy ester 17 could be induced by performing the reaction under microwave irradiation at 140 °C in the presence of excess of lithium *tert*butylthiolate (Scheme 9). The resulting carboxylic acid 18 was converted to the target ligand 10b-syn following the same procedure used for the preparation of 10a-syn.

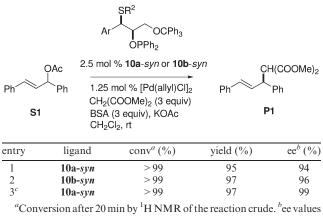
Ligands **10a**-syn and **10b**-syn were next evaluated in AAA of **S1** (Table 5).

Again, very high activities were observed for both ligands, and complete conversions were recorded after 20 min at rt. Interestingly, a positive effect on the enantioselectivity was observed with the *syn*-type ligands, **P1** being obtained in 94% and 96% ee with **10a**-*syn* and **10b**-*syn*, respectively. For comparison, the corresponding ligands with *anti*-configuraSCHEME 9. Synthesis of Ligand 10b-syn^a



^{*a*}Reagents and conditions: (a) *t*-BuSH (3 equiv), LiOH (3 equiv), dioxane/water, microwave irradiation 140 °C, 3 h; (b) $BH_3 \cdot SMe_2$ (4 equiv), THF, 16 h; (c) CICPh₃ (1.2 equiv), pyridine, 90 °C, 18 h; (d) CIPPh₂ (1.1 equiv), NEt₃ (1.2 equiv), 10 mol % of 4-DMAP, toluene, rt, 40 min.

 $\label{eq:configuration} TABLE \, 5. \qquad Effect \, of \, the \, Relative \, Configuration \, (C2/C3) \, on \, Conversion \\ and \, Enantioselectivity$



by chiral HPLC. ^c0 °C, 240 min.

tion (**10a** and **10b**, Scheme 5) afforded the alkylation product with 86% and 91% ee, respectively. Very gratifyingly, enantioselectivity increased to 99% with ligand **10a**-*syn* by simply lowering the reaction temperature to 0 °C, although a somewhat longer reaction time (240 min) was required to achieve high conversion under these conditions (entry 3).

The synthesis of **10a**-syn and **10b**-syn completed the modular optimization of the new family of *P*,*S*-ligands. Along this way, enantioselectivities in the archetypal substitution of 1,3-diphenylallyl acetate with dimethyl malonate have increased from 10 to 12% with **4a**-**d** to 99% with **10a**-syn, while high catalytic activities were observed over the whole series.

To compare the merits of the ligands resulting from the structural optimization with other well-established P,Sligands, we have collected in Figure 5 the enantioselectivities in the reaction of S1 with dimethyl malonate leading to P1 and the corresponding reaction conditions for full conversion. It is interesting to note that the levels of catalytic activity and enantiocontrol achieved with one of our optimized ligands (10b-syn) are among the highest recorded in the considered ligand set. In particular, it is worth noting that 10b-syn induces very high enantioselectivity at room temperature, where the alkylation is completed in very short reaction times.

As a final aspect of this research, the behavior of the optimal ligand set (**10a**,**b** and **10a**,**b**-*syn*) was studied in the alkylation of the more challenging substrates **S2** and **S3** (Table 6) and in

⁽²⁵⁾ Ramón, R.; Alonso, M.; Riera, A. Tetrahedron: Asymmetry 2007, 18, 2797.

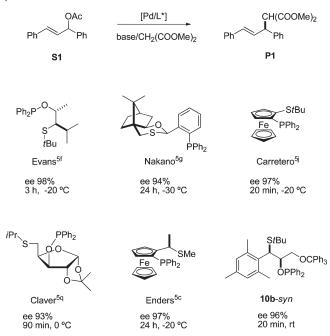


FIGURE 5. Behavior of representative *P*,*S*-ligands reported in the literature in the Pd-catalyzed allylic alkylation of 1,3-diphenylpropenyl acetate with dimethyl malonate.

 $\label{eq:constraint} \begin{array}{ll} TABLE\,6. & Enantioselective Allylic Alkylation for Substrates\,S2\,and\,S3\\ with Dimethyl Malonate \end{array}$

	Ar	² OCPh ₃ DPPh ₂	Ar OPPh	OCPh ₃ 2			
Ph	Ph OAc 2.5 mol % 10a ,10b and syn isomers Ph CH(CO ₂ Me) ₂						
Ph	, − [−]	1.25 mol % [Pc	d(allyl)Cl]2	Ph	`R		
		CH ₂ (COOMe) ₂	(3 equiv)				
S2 : R	= Ph	BSA (3 equiv),		P2 : R =	Ph		
S3 : R	= Me	CH ₂ Cl ₂ , rt		P3 : R =	Me		
entry	ligand	substrate	time (h)	$\operatorname{conv}^{a}(\%)$	ee^{b} (%)		
1	10a	S2	16	54	97		
2	10b	S2	16	46	96		
3	10a-syn	S2	5	93	40		
4	10b-syn	S2	5	92	70		
5	10a -	S 3	16	45	79		
6	10b	S 3	16	30	82		
7	10a-syn	S 3	5	99	4		
8	10b-syn	S 3	5	99	20		
^{<i>a</i>} Conversion by ¹ H NMR of the reaction crude. ^{<i>b</i>} ee values by chiral HPLC.							

the substitution of **S1** with more challenging nucleophiles such as benzylamine and benzyl alcohol (Table 7).

Not unexpectedly (see Table 6),²⁶ these trisubstituted π -allyl precursors proved to be much less reactive in front of alkylation than the 1,3-diphenyl disubstituted substrate **S1**.

At room temperature, complete conversion of S2 and S3 was observed over a period of 5 h using ligands 10a,b-syn. Unfortunately, these ligands, which proved to be the most

 TABLE 7.
 Effect of the Nucleophile on Conversion and Enantioselectivity

	Ph S1	PAC [Pd	-Ligands (allyI)CI]₂ CH₂CI₂, rt	Nu Ph Ph P6: NuH=BnNH ₂ P7: NuH=BnOH	
entry	ligand	NuH	time (h)	yield (%)	ee^{a} (%)
1	10a	BnNH ₂ ^b	2	98	75
2	10b	$BnNH_{2}^{b}$	16	73	73
3	10a-syn	$BnNH_2^{-b}$	16	65	92
4	10b-syn	$BnNH_2^{b}$	4	97	94
5	10b-syn	$BnNH_2^c$	10	95	95
6	10a-syn	$BnOH^d$	3	98	90
7	10b-syn	$BnOH^d$	3	96	93
8	10b-syn	BnOH ^e	16	55	86
9	10b-syn	BnOH	3	96	94
a_{a} and b_{a} by this 1 UDL C b_{a} be still and this set 1.25 and 1.0/ of					

^{*a*}ee values by chiral HPLC. ^{*b*}Reaction conditions: 1.25 mol % of [Pd(η^3 -C₃H₅)Cl]₂, 3.75 mol % of *P*,*S*-ligand, benzylamine (3 equiv), CH₂Cl₂, rt. ^{*c*}Reaction at 0 °C. ^{*d*}Reaction conditions: 2 mol % of [Pd-(η^3 -C₃H₅)Cl]₂, 4.2 mol % of *P*,*S*-ligand, benzyl alcohol (3 equiv), Cs₂CO₃ (3 equiv), CH₂Cl₂, rt. ^{*e*}Reaction at 0 °C in CH₂Cl₂. ^{*f*}Reaction at 0 °C in toluene.

effective ones for the allylic substitution of S1, afforded the alkylation product in much lower enantioselectivities (entries 3, 4, 7, and 8). Presumably, this lack of enantioselectivity is derived from a different ligand structural requirement for the successful alkylation of trisubstituted π -allyl precurors compared to the disubstituted substrate S1. On the other hand, using ligands 10a and 10b under standard conditions for AAA, excellent enantioselectivities were observed for substrate S2 (96% and 97% ee) and good asymmetric inductions for substrate S3 (79% and 82% ee; entries 1, 2, 5 and 6). Interestingly, for substrate S2 and using ligand 10a, when the reaction was stopped at 54% conversion, the unreacted allylic substrate was recovered in 70% ee (entry 1), which represents a rather high selectivity for one of the substrate enantiomers.

Finally, the optimal set of ligands was also examined in the palladium-catalyzed reaction of 1,3-diphenylpropenyl acetate (S1) with different heteroatom nucleophiles. Benzylamine and benzyl alcohol were chosen as representative models for N-nucleophiles and O-nucleophiles, respectively, and the results obtained in these reactions are summarized in Table 7.

The catalytic system involving ligand **10b**-syn was found to be the most effective for the amination and etherification of S1. After 4 h at rt, the amination product P6 was obtained in 97% yield and 94% ee (entry 4), whereas the analogous ligand 10b afforded lower yield and enantioselectivity after 16 h (entry 2). Enantioselectivity increased slightly by lowering the temperature to 0 °C, but longer reaction time was required to achieve complete conversion (entry 5). Furthermore, using benzyl alcohol as a nucleophile, **10b**-syn was the most effective ligand for this asymmetric etherification reaction, affording the product **P7** in 96% yield and 93% ee after 3 h at rt (entry 7). The same reaction at 0 °C afforded the product in lower selectivity (86% ee) and lower yield (55% yield) after 16 h (entry 8). When the solvent was changed to toluene, the product was obtained in almost quantitative yield (96%) after 3 h with high enantioselectivity (94% ee)(entry 9). While numerous examples of enantioselective allylic amination have been reported, successful examples

^{(26) (}a) Auburn, P. R.; Mackenzie, P. B.; Bosnich, B. J. Am. Chem. Soc.
1985, 107, 2033. (b) Dawson, G. J.; Williams, J. M. J.; Coote, S. J. Tetrahedron: Asymmetry 1995, 6, 2535. (c) Dawson, G. J.; Williams, J. M. J.; Coote, S. J. Tetrahedron Lett. 1995, 36, 461.

of asymmetric allylic etherification are very scarce. Only a recent account by Chan et al. reports high enantioselectivities in this process.^{5t} When both catalytic systems are compared, it is worth mentioning that **10b**-*syn* behaves as a more active catalyst in the considered reaction while leading to similar levels of asymmetric induction.

Structural Studies. The binding mode of ligand **10a**-syn was established by X-ray crystallography. A suitable crystal of the PdCl₂ complex of **10a**-syn was obtained from dichloromethane/diethyl ether. Complex (**10a**-syn) · **PdCl₂** adopts a twist-boat geometry, and only one epimer of the adamantylthio substituent is observed in the crystal. In this epimer, the adamantyl group is in an *anti*-orientation with respect to the adjacent phenyl group. Furthermore, ¹H NMR and ³¹P NMR studies on (**10a**-syn) · **PdCl₂** in CD₂Cl₂ at rt also showed only one diastereometric complex, thus indicating that the sulfur inversion is blocked. In addition, the larger *trans* influence of the phosphorus atom (Pd-Cl₁ = 2.359 Å) compared to sulfur (Pd-Cl₂ = 2.312 Å).

Interestingly, a complementary X-ray diffraction study performed on 4m·PdCl₂ showed that changing the relative configuration of the carbons bearing the heteroatoms (C2/C3) in the ligand backbone leads to a conformational change of the 6-membered ring Pd-chelate, in agreement with previous observations by Evans.^{5f} While complex (10a-syn)· PdCl₂ showed a twist-boat conformation, crystals of 4m·PdCl₂ grown in dichloromethane/diethyl ether showed a half-chair conformation in the corresponding six-membered chelate ring (see Figures 1 and 2 in the Supporting Information).²⁷

Origin of the Enantioselectivity. It has been generally accepted that the enantiodifferentation step in the Pd-catalyzed allylation is the substitution of π -allyl complexes with nucleophiles. Two bonds in a *trans* position compete for one metal *d*-orbital for a π -back-bond. The stronger the π -acceptor ability of the ligand is, the more electronic density is removed from the metal *d*-orbital. Through this mechanism, the electronic density at one of the allylic carbon atoms is strongly reduced by a π -acceptor in a *trans* position. Accordingly, nucleophilic attack should occur predominantly at the allyl terminus located *trans* to the best π -acceptor (P > S).

Since the (*S*) enantiomer of the alkylation product is predominantly obtained, the reaction must proceed through an M-type rather than W-type intermediate (Figure 6, top). It can be reasoned from a qualitative perspective that the M-type complex would react faster than its diastereomer W-type complex. Thus, attack of the nucleophile to the allyl

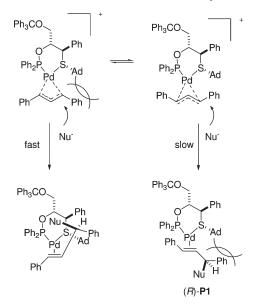


FIGURE 6. Proposed transition-state models for the asymmetric allylic substitutions.

terminus *trans* to phosphorus in the M-type complex will lead to a release of steric congestion between the S-adamantyl substituent and the proximal phenyl substituent on the allyl unit, while attack to the corresponding carbon in the W-type complex will be accompanied by an increasing steric interaction between the S-adamantyl group and the corresponding phenyl group on the allyl moiety as the rotation from the starting π -allyl complex to the final Pd(0)-olefin product (*R*)-**P1** proceeds (Figure 6).

Conclusions

In summary, a library of modularly constructed phosphinite thioether ligands derived from the ring-opening of enantiopure epoxides has been synthesized and evaluated in the Pd-catalyzed allylic substitution reactions of several substrates. These P,S-ligands have been successfully optimized by following a modular approach. Over this process, the enantioselectivity recorded in the benchmark reaction has evolved from ca. 10% in the initial ligands to 99% in the optimized structures. In general, activities and enantioselectivities are mainly controlled by the steric interactions between the substituent on sulfur, and the skeletal aryl group, and by the relative stereochemistry of the stereocenters at C2 and C3 on the ligand skeleton. Excellent enantioselectivities were obtained in the reaction of rac-1,3-diphenyl-2-propenylacetate with dimethyl malonate (up to 99%), benzylamine (up to 95%), and a much less common O-nucleophile, such as benzyl alcohol (up to 94%). The development of further applications of these ligands is now in progress in our laboratories.

Experimental Section

General Procedure for the Ring-Opening of Epoxides 2a-d by Thiols. To a solution of the epoxide (1 mmol) and sodium hydroxide (2 mmol) in dioxane/water (10:1 v/v) was added the corresponding thiol (2 mmol, 2 equiv). The mixture was heated at the indicated temperature, and reaction progress was monitored by TLC until disappearance of the starting epoxide

⁽²⁷⁾ Crystal data for **4** m·PdCl₂ at 100 K: C₂₆H₃₁Cl₂O₂P₁Pd₁S₁ × 2 (the unit cell contains two independent molecules), 615.84 gmol⁻¹, triclinic, *P*1, *a* = 10.6012(5) Å, *b* = 12.3725(5) Å, *c* 12.9208(6) Å, *a* = 67.7560(10)°, β = 65.9300(10)°, γ = 65.8490(10)°, *V* = 1365.60(11) Å³, *Z* = 2, ρ_{calcd} = 1.498 Mg/m³, R_{1obs} = 0.0230 (R_{1ref} = 0.0243), W_{2bob} = 0.0554 (W_{2ref} = 0.0561), for 16893 reflections with *I* > 2 σ (*I*) (for 17508 reflections (R_{int} 0.0227) with a total of 28395 measured reflections), Flack (std): -0.028(8), diffracting 2 θ range: 3.58-38.13°, goodness-of-fit on *F*² = 1.015, largest diff peak (hole) = 1.437 (-0.607) e Å⁻³. CCDC: 755742. Crystal data for (**10***a*-*syn*)·**PdCl₂ at** 100 K: C₅₀H₄₀Cl₂O₂P₁Pd₁S₁, 922.22 gmol⁻¹, orthorhombic, *P*₂L₂1, *a* = 11.2090(5) Å, *b* = 12.4178(6) Å, *c* 31.5636(15) Å, *V* = 4393.4(4) Å³, *Z* = 4, ρ_{calcd} = 1.394 Mg/m³, R_{1obs} = 0.0215 (R_{1ref} = 0.0223), w_{2obs} = 0.0547 (w_{2ref} = 0.0550), for 24842 reflections with *I* > 2 σ (*I*) (for 25376 reflections) (R_{int} : 0.0262) with a total measured of 89664 reflections), Flack (std): -0.013(6), diffracting 2 θ range: 2.77–39.60°, goodness-of-fit on *F*² = 1.105, largest diff peak (hole) = 1.539 (-0.456) e Å⁻³.CCDC: 755743. The supplementary crystallographic data for this paper can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.

(ca. 20–90 min). The mixture was allowed to reach room temperature. Then, 10 mL of water was added, and the mixture was extracted with CH_2Cl_2 (3 × 15 mL). The combined organic extracts were dried over Na_2SO_4 . The solvent was removed under vacuum, and the crude product was purified by flash chromatography on silica gel, eluting with hexane/ethyl acetate mixtures to give the desired product.

(1R,2S)-3-Methoxy-1-phenyl-1-(phenylthio)propan-2-ol (3a). The following reagents were combined in the amounts indicated below, and submitted to the general procedure for the thiolate ring-opening of epoxides, by performing the reaction at 65 °C for 20 min: 2a (130 mg, 0.79 mmol), thiophenol (165 μL, 1.58 mmol), and NaOH (63 mg, 1.58 mmol). The product was purified by flash chromatography (90% hexane, 10% ethyl acetate) to yield 3a as a white solid (200 mg, 92% yield): mp 90–91 °C; $[\alpha]^{26}_{D}$ –179.4 (*c* 1.15, CHCl₃); IR (neat) 3372, 2891, 1583, 1493, 1481, 1453, 1438, 1348, 1189, 1091, 1073, 1059, 942, 736, 705, 687, 567 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.19–7.37 (m, 10H), 4.35 (d, ${}^{3}J = 6.19$ Hz, 1H), 4.12–4.17 (m, 1H), 3.40-3.47 (m, 2H), 3.30 (s, 3H), 2.47 (d, ${}^{3}J = 4.27$ Hz, 1H, OH); ¹³C NMR (100 MHz, CDCl₃) δ 138.2, 134.2, 132.3, 128.9, 128.8, 128.4, 127.6, 127.4, 73.8, 71.9, 59.0, 56.2; HRMS (ES+) m/z calcd for C₁₆H₁₈O₂SNa: 297.0925 [M + Na]⁺, found 297.0936.

(1*R*,2*S*)-3-(Benzyloxy)-1-phenyl-1-(phenylthio)propan-2-ol (3b). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 65 °C for 60 min: 2b (111 mg, 0.46 mmol), thiophenol (96 μL, 0.92 mmol), and NaOH (37 mg, 0.92 mmol). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield 3b as a colorless oil (117 mg, 72% yield): $[\alpha]^{26}_{D}$ –148.7 (*c* 0.58, CHCl₃); IR (neat) 3422, 3058, 3027, 2908, 2861, 1700, 1599, 1582, 1493, 1479, 1204, 1087, 1067, 1025, 736, 693 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.19–7.35 (m, 15H), 4.48 (d, ²*J* = 11.8 Hz, 1H), 4.45 (d, ²*J* = 11.8 Hz, 1H), 4.37 (d, ³*J* = 6.15 Hz, 1H), 4.16–4.22 (m, 1H), 3.55 (dxd, ²*J* = 9.6 Hz, ³*J* = 2.1 Hz, 1H), 3.54 (dxd, ²*J* = 9.6 Hz, ³*J* = 3.3 Hz, 1H), 2.51 (d, ³*J* = 4.6 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 138.3, 137.8, 134.3, 132.3, 128.9, 128.4, 127.8, 127.7, 127.6, 127.3, 73.4, 72.0, 71.6, 56.3; HRMS (ES+) *m*/*z* calcd for C₂₂H₂₂O₂NaS 373.1238 [M + Na]⁺, found 373.1247.

(1*R*,2*S*)-3-(Benzhydryloxy)-1-phenyl-1-(phenylthio)propan-2ol (3c). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 65 °C for 120 min: 2c (250 mg, 0.79 mmol), thiophenol (165 μL, 1.58 mmol), and NaOH (63 mg, 1.58 mmol). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield 3c as a white solid (305 mg, 90% yield): mp 74–75 °C; $[\alpha]^{26}_D$ –117.9 (*c* 0.51, CHCl₃); IR (neat) 3547, 3057, 3025, 2866, 2854, 1596, 1582, 1492, 1470, 1449, 1338, 1099, 1075, 965, 750, 737, 691, 612 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.18–7.33 (m, 20H), 5.3 (s, 1H), 4.41 (d, ³*J* = 6.07 Hz, 1H), 4.20–4.25 (m, 1H), 3.54 (d, ³*J* = 5.2 Hz, 2H), 2.51 (d, ³*J* = 5.01 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 141.7, 138.3, 134.5, 132.2, 128.9, 128.8, 128.4, 128.4, 128.3, 127.6, 127.5, 127.3, 126.9, 126.9, 84.3, 72.1, 70.4, 56.4; HRMS (ES+) *m/z* calcd for C₃₂H₃₄O₂NaS 505.2177 [M + Na]⁺, found 505.2172.

(1*R*,2*S*)-1-Phenyl-1-(phenylthio)-3-(trityloxy)propan-2-ol (3d). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 65 °C for 120 min: 2d (310 mg, 0.79 mmol), thiophenol (165 μ L, 3.95 mmol, 5 equiv), and NaOH (158 mg, 3.95 mmol, 5 equiv). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield 3d as a white solid (293 mg, 74% yield): mp 91–92 °C; [α]²⁶_D –70.0 (*c* 0.65, CHCl₃); IR (neat) 3424, 3057, 3026, 2922, 2851, 1596, 1582, 1490, 1448, 1069, 1027, 899,743, 695, 632 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.36–7.39 (m, 6H), 7.18–7.29 (m, 19 H), 4.44 (d,

 ${}^{3}J = 5.7$ Hz, 1H), 4.11–4.17 (m, 1H), 3.28 (dxd, ${}^{2}J = 9.53$ Hz, ${}^{3}J = 5.76$ Hz, 1H), 3.18 (dxd, ${}^{2}J = 9.53$ Hz, ${}^{3}J = 5.40$ Hz, 1H), 2.39 (d, ${}^{3}J = 5.12$ Hz, 1H); ${}^{13}C$ NMR (100 MHz, CDCl₃) δ 143.7, 137.9, 134.6, 131.9, 128.9, 128.8, 128.6, 128.3, 127.8, 127.4, 127.2, 127.1, 87.0, 72.3, 64.9, 56.4; HRMS (ES+) m/z calcd for C₃₄H₃₀O₂NaS 525.1864 [M + Na]⁺, found 525.1857.

(1R,2S)-1-(3,5-Dimethylphenylthio)-3-methoxy-1-phenylpropan-2-ol (3e). The following reagents were combined in the amounts indicated according to the general procedure for the thiolate ring-opening of epoxides and heated at 65 °C for 20 min: 2a (177 mg, 1.078 mmol), 3,5-dimethylbenzenethiol (326 µL, 2.156 mmol), and NaOH (86.2 mg, 2.156 mmol). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield 3e as an oil (270 mg, 83% yield): $[\alpha]_{D}^{26}$ -204.4 (*c* 0.7, CHCl₃); IR (neat) 3438, 2914, 1599, 1579, 1491, 1451, 1190, 1120, 1074, 962, 847, 748, 699, 686 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.24-7.38 (m, 5H), 6.97 (s, 2H), 6.85 (s, 1H), 4.33 (d, ${}^{3}J = 5.85$ Hz, 1H), 4.11–4.16 (m, 1H), 3.45 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{2}J = 9.6 Hz, ${}^{3}J = 4.1$ Hz, 1H), 3.4 (dxd, {}^{3}J = 4.1 6.1 Hz, 1H), 3.3 (s, 3H), 2.56 (d, ${}^{3}J = 4.1$ Hz, 1H, OH), 2.24 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 138.5, 138.3, 133.5, 129.9, 129.3, 128.8, 128.4, 127.5, 73.9, 71.7, 59.0, 56.1, 21.1; HRMS (ES+) m/z calcd for C₁₈H₂₂O₂NaS 325.1238 [M + Na]⁺, found 325.1222

(1*R*,2*S*)-1-(2,6-Dimethylphenylthio)-3-methoxy-1-phenylpropan-2-ol (3f). The following reagents were combined in the amounts indicated according to the general procedure fo the thiolate ring-opening of epoxides and heated at 65 °C for 40 min: 2a (177 mg, 1.078 mmol), 2,6-dimethylbenzenethiol (326 μL, 2.156 mmol), and NaOH (86.2 mg, 2.156 mmol). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield 3f as a white solid (227 mg, 70% yield): mp 83–84 °C; $[\alpha]^{26}_{D}$ –248.9 (*c* 0.61, CHCl₃); IR (neat) 3457, 2918, 2898, 2828, 2820, 1493, 1457, 1437, 1078, 1063, 1030, 876, 775, 731, 699 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.19–7.25 (m, 5H), 7.01–7.1 (m, 3H), 4.14–4.19 (m, 1H), 3.96 (d, ³J = 6.7 Hz, 1H), 3.51 (dxd, ²J = 9.6 Hz, ³J = 3.8 Hz, 1H), 3.46 (dxd, ²J = 9.6 Hz, ³J = 6.1 Hz, 1H), 3.31 (s, 3H), 2.44 (d, ³J = 4.1 Hz, 1H; OH), 2.36 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 143.5, 138.8, 132.0, 128.6, 128.5, 128.3, 128.1, 127.5, 74.2, 72.1, 59.0, 55.7, 21.9; HRMS (ES+) *m*/*z* calcd for C₁₈H₂₂O₂NaS 325.1238 [M + Na]⁺, found 325.1231.

(1R,2S)-3-Methoxy-1-(4-methoxyphenylthio)-1-phenylpropan-2-ol (3g). The following reagents were combined in the amounts indicated according to the general procedure for the thiolate ringopening of epoxides and heated at 65 °C for 20 min: 2a (115 mg, 0.7 mmol), 4-methoxybenzenethiol (178 µL, 1.4 mmol), and NaOH (56 mg, 1.4 mmol). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield **3g** as an oil (190 mg, 89% yield): $[\alpha]_{D}^{26}$ -202.8 (*c* 0.77, CHCl₃); IR (neat) 3434, 3060, 2925, 2895, 2835, 1591, 1570, 1492, 1452, 1284, 1243, 1173, 1120, 1028, 1120, 1028, 960, 827, 699, 640 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.21-7.29 (m, 7H), 6.76-6.78 (br t, 1H), 6.74-6.75 (br t, 1H), 4.16 (d, ${}^{3}J = 6.2$ Hz, 1H), 4.09-4.14 (m, 1H), 3.76 (s, 3H), 3.47 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 3.8$ Hz, 1H), 3.42 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 6.14$ Hz, 1H), 3.3 (s, 3H), 2.53 (d, ${}^{3}J = 3$.8 Hz, 1H; OH); ${}^{13}C$ NMR (100 MHz, CDCl₃) δ 159.7, 138.3, 135.7, 128.9, 128.4, 127.5, 124.1, 114.5, 74.1, 71.4, 59.1, 57.5, 55.3; HRMS (ES+) m/z calcd for C₁₇H₂₀O₃NaS: 327.1031 [M + Na]⁺, found 327.1044.

(1*R*,2*S*)-1-(4-Bromophenylthio)-3-methoxy-1-phenylpropan-2ol (3h). The following reagents were combined in the amounts indicated according to the general procedure for the thiolate ring-opening of epoxides and heated at 65 °C for 30 min: 2a (200 mg, 1.218 mmol), 4-bromobenzenethiol (485 mg, 2.436 mmol), and NaOH (97.4 mg, 2.436 mmol). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield 3h as a white solid (350 mg, 81% yield): mp 70–71 °C; $[\alpha]^{26}_{\rm D}$ –207.9 (c 0.81, CHCl₃); IR (neat) 3387, 2942, 2904, 2841, 1491, 1468, 1452, 1384, 1371, 1260, 1189, 1091, 1079, 1062, 1007, 994, 944, 875, 817, 790, 732, 727, 698 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.23–7.35 (m, 7H), 7.16 (br s, 1H), 7.14 (br s, 1H), 4.32 (d, ³J = 6.14 Hz, 1H), 4.14 (br m, 1H), 3.39–3.46 (m, 2H), 3.3 (s, 3H), 2.46 (br s, 1H, OH); ¹³C NMR (100 MHz, CDCl₃) δ 137.9, 133.6, 133.5, 131.9, 128.8, 128.5, 127.8, 121.5, 73.7, 72.0, 59.0, 56.2; HRMS (ES+) *m/z* calcd for C₁₆H₁₇O₂NaSBr 375.0030 [M + Na]⁺, found 375.0042.

(1*R*,2*S*)-1-(4-*tert*-Butylphenylthio)-3-methoxy-1-phenylpropan-2-ol (3i). The following reagents were combined in the amounts indicated according to the general procedure for the thiolate ringopening of epoxides and heated at 65 °C for 20 min: 2a (200 mg, 1.218 mmol), 4-*tert*-butylbenzenethiol (433 μL, 2.436 mmol), and NaOH (97.4 mg, 2.436 mmol). The product was purified by flash chromatography (90% hexane, 10% ethyl acetate) to yield 3i as a white solid (342 mg, 85% yield): mp 71–72 °C; $[\alpha]^{27}_D$ –190.5 (*c* 0.63, CHCl₃); IR (neat) 3363, 2951, 2866, 2833, 1493, 1358, 1270, 1195, 1107, 1093, 916, 878, 810, 703, 683 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.24–7.37 (m, 9H), 4.3 (d, ³*J* = 6.7 Hz, 1H), 4.09–4.14 (m, 1H), 3.42 (dxd, ²*J* = 9.6 Hz, ³*J* = 4.1 Hz, 1H), 3.38 (dxd, ²*J* = 9.6 Hz, ³*J* = 6 Hz, 1H), 3.28 (s, 3H), 2.56 (d, ³*J* = 4.1 Hz, OH, 1H), 1.27 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 150.83, 138.3, 132.29, 130.61, 128.84, 128.4, 127.6, 126.0, 73.9, 71.7, 59.0, 56.5, 34.5, 31.2; HRMS (ES+) *m*/*z* calcd for C₂₀H₂₆-O₂S 330.1654 [M], found 330.1656.

(1R,2S)-3-Methoxy-1-(naphthalen-2-ylthio)-1-phenylpropan-2-ol (3j). The following reagents were combined in the amounts indicated according to the general procedure for the thiolate ring-opening of epoxides and heated at 65 °C for 20 min: 2a (200 mg, 1.218 mmol), 2-naphthalenethiol (394 mg, 2.436 mmol), and NaOH (97.4 mg, 2.436 mmol). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield 3j as a white solid (352 mg, 89% yield): mp $85-86 \text{ °C}; [\alpha]^{27} \text{ }_{\text{D}} -247.3 (c 0.49, \text{CHCl}_3); \text{ IR (neat) } 3443, 2924,$ 2900, 1493, 1469, 1454, 1121, 1105, 1078, 1063, 990, 942, 866, 816, 751, 730, 696 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.79 (br, 1H), 7.74-7.77 (m, 1H), 7.67-7.70 (m, 2H), 7.38-7.47 (m, J = 9.67 Hz, ${}^{3}J = 5.9$ Hz, 1H), 3.28 (s, 3H), 2.53 (d, ${}^{3}J =$ 4.4 Hz, 1H; OH); ¹³C NMR (100 MHz, CDCl₃) δ 138.2, 133.5, 132.3, 131.7, 130.9, 129.4, 128.8, 128.5, 128.4, 127.7, 127.6, 127.3, 126.4, 126.2, 73.8, 72.0, 59.0, 56.1; HRMS (ES+) m/z calcd for $C_{20}H_{20}O_2NaS$ 347.1082 [M + Na]⁺, found 347.1085.

(1R,2S)-1-(isopropylthio)-3-methoxy-1-phenylpropan-2-ol (3k). The following reagents were combined in the amounts indicated according to the general procedure for the thiolate ring-opening of epoxides and heated at 65 °C for 30 min: 2a (215 mg, 1.31 mmol), 2-propanethiol (250 µL, 2.62 mmol) and NaOH (105 mg, 2.62 mmol). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield 3k as a white solid (253 mg, 83% yield). Mp 72–73 °C; $[\alpha]^{26}$ – 172.1 (*c* 0.6, CHCl₃); IR (neat) 3438, 2958, 2890, 2862, 2837, 1490, 1462, 1453, 1444, 1246, 1152, 1120, 1062, 1050, 992, 942, 749, 700 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) & 7.38-7.41 (m, 2H), 7.31-7.35 (m, 2H), 7.24-7.28 (m, 1H), 4.04-4.09 (m, 1H), 4.02 (d, ${}^{3}J = 6.4$ Hz, 1H), 3.48 (dxd, ${}^{2}J =$ 9.6 Hz, ${}^{3}J = 3.8$ Hz, 1H), 3.42 (dxd, ${}^{2}J = 9.6$ Hz, ${}^{3}J = 6.3$ Hz, 1H), 3.36 (s, 3H), 2.66 (sept, ${}^{3}J = 6.7$ Hz, 1H), 2.41 (br s, 1H, OH), 1.24 (d, ${}^{3}J = 6.7$ Hz, 3H), 1.16 (d, ${}^{3}J = 6.7$ Hz, 3H); ${}^{13}C$ NMR (100 MHz, CDCl₃) δ 139.5, 128.8, 128.5, 127.4, 74.1, 72.7, 59.1, 51.6, 34.7, 23.5, 23.2; HRMS (ES+) m/z calcd for C₁₃H₂₀O₂NaS: $263.1082 [M + Na]^+$, found 263.1084.

(1R,2S)-1-(c yclohexylthio)-3-methoxy-1-phenylpropan-2-ol (3). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 65 °C for 20 min: 2a (146 mg, 0.89 mmol), cyclohexylthiol (224 μ L, 1.78 mmol) and NaOH (71 mg, 1.78 mmol). The product was purified by flash chromatography (90% hexane,10%)

ethyl acetate) to yield **3I** as a white solid (180 mg, 72% yield). Mp 68–69 °C; $[\alpha]^{26}_{\rm D}$ –178.4 (*c* 0.84, CHCl₃); IR (neat) 3422, 3056, 3028, 2915, 2892, 2847, 2812, 1491, 1449, 1111, 1095, 1074, 1061, 981, 946, 698, 680 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.23–7.39 (m, 5H), 4.03–4.09 (m, 2H), 3.46 (dxd, ²J = 9.6 Hz, ³J = 3.6 Hz, 1H), 3.41 (dxd, ²J = 9.6 Hz, ³J = 5.8 Hz, 1H), 3.35 (s, 3H), 2.43–2.50 (m, 2H), 1.93–1.97 (br d, 1H), 1.54–1.80 (m, 4H), 1.13–1.38 (m, 5H); ¹³C NMR (100 MHz, CDCl₃) δ 139.6, 128.7, 128.4, 127.4, 74.1, 72.7, 59.1, 51.1, 43.3, 33.7, 33.5, 25.9, 25.8, 25.7; HRMS (ES+) *m/z* calcd for C₁₆H₂₄O₂NaS: 303.1395 [M + Na]⁺, found 303.1377.

(1R,2S)-1-(tert-butylthio)-3-methoxy-1-phenylpropan-2-ol (3m). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 75 °C for 1 h: 2a (110 mg, 0.67 mmol), tertbutylthiol (148 µL, 1.34 mmol) and NaOH (53.6 mg, 1.34 mmol). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield **3m** as a white solid (115 mg, 71% yield). Mp 74–75 °C; $[\alpha]_{D}^{26}$ –179.9 (c 0.75, CHCl₃); IR (neat) 3445, 2962, 2937, 2908, 2860, 2819, 1601, 1490, 1453, 1363, 1264, 1191, 1157, 1117, 1100, 1076, 1062, 986, 940, 743, 698 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.39-7.42 (m, 2H), 7.29-7.33 (m, 2H), 7.22-7.26 (m, 1H), 3.99-4.05 (m, 2H), 3.36-3.43 (m, 2H), 3.34 (s, 3H), 2.42 (d, ${}^{3}J$ = 4.68, 1H, OH), 1.25 (s, 9H); ${}^{13}C$ NMR (100 MHz, CDCl₃) & 141.2, 128.8, 128.4, 127.1, 73.8, 73.6, 59.0, 50.7, 44.2, 31.4; HRMS (ES+) m/z calcd for C₁₄H₂₂O₂NaS: 277.1238 [M + Na]⁺, found 277.1233.

(1R,2S)-1-(adamantylthio)-3-methoxy-1-phenylpropan-2-ol (3n). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 80 °C for 60 min: 2a (61 mg, 0.37 mmol), adamanthanethiol (263 mg, 1.48 mmol, 4 equiv) and NaOH (59.4 mg, 1.48 mmol, 4 equiv). The product was purified by flash chromatography (90% hexane,10% ethyl acetate) to yield 3n as a white solid (80 mg, 64% yield). Mp 73-74 °C; $[\alpha]_{D}^{26}$ -176.8 (c 0.96, CHCl₃); IR (neat) 3443, 3058, 3025, 2901, 2847, 1599, 1491, 1450, 1342, 1299, 1189, 1120, 1042, 963, 747, 698, 666 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.39–7.41 (m, 2H), 7.28–7.32 (m, 2H), 7.21-7.26 (m, 1H), 4.11 (d, ${}^{3}J = 5.97$ Hz, 1H), 3.98-4.03 (m, 1H), 3.35-3.41 (m. 2H), 3.32 (s, 3H), 2.50 (d, ${}^{3}J = 4.68$, 1H, OH), 1.98 (br s, 3H), 1.75–1.82 (br m, 6H), 1.59–1.67 (br m, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 141.6, 128.7, 128.3, 127.0, 73.9, 73.7, 59.0, 48.1, 46.5, 43.8, 36.2, 29.7; HRMS (ES+) m/z calcd for $C_{20}H_{28}O_2NaS: 355.1708 [M + Na]^+$, found 355.1703.

(2*S*,3*R*)-3-(adamantylthio)-3-pherylpropane-1,2-diol (8a). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 80 °C for 120 min: 1a (200 mg, 1.33 mmol), adamanthanethiol (475 mg, 2.66 mmol) and NaOH (106 mg, 2.66 mmol). The product was purified by flash chromatography (80% hexane, 20% ethyl acetate) to yield 8a as a white solid (250 mg, 59% yield). Mp 69–70 °C; $[\alpha]^{26}_{D}$ –201.9 (*c* 0.9, CHCl₃); IR (neat) 3402, 2901, 2847, 1598, 1498, 1449, 1341, 1300, 1076, 1042, 699 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.22–7.42 (m, 5H), 4.11 (d, ³*J* = 6.9 Hz, 1H), 3.86–3.91 (m, 1H), 3.64–3.7 (m. 1H), 3.55–3.6 (m, 1H), 2.37(d, ³*J* = 5.83 Hz, 1H, OH), 2.04–2.07 (m, 1H, OH), 1.98 (br s, 3H), 1.73–1.85 (br m, 6H), 1.59–1.67 (br m, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 141.5, 128.6, 128.6, 127.2, 74.9, 64.0, 48.2, 46.7, 43.8, 36.1, 29.7; HRMS (ES+) *m/z* calcd for C₁₉H₂₇O₂S: 319.1732 [M + H]⁺, found 319.1726.

Selective Protection of the Primary Hydroxyl Group in 8a as a Trityl Ether. (1*R*,2*S*)-1-phenyl-1-(adamantylthio)-3-(trityloxy)-propan-2-ol (9a). A solution of 8a (235 mg, 0.74 mmol) and triphenylmethyl chloride (252 mg, 0.89 mmol) in pyridine (6 mL) was heated at 90 °C for 18 h under N₂. The solvent was removed *in vacuo*, and the residual oil was chromatographed using hexane:Et₂O (9:1) as eluent to give 283 mg (68%) of 9a as a white solid. Mp 77–78 °C; $[\alpha]^{26}_{D}$ –79.5 (*c* 0.9, CHCl₃); IR (neat)

3448, 3083, 3056, 3023, 2901, 2847, 1597, 1490, 1448, 1342, 1299, 1218, 1072, 1042, 1031, 761, 745, 632 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.40–7.43 (m, 6H), 7.14–7.29 (m, 14H), 4.2 (d, ³*J* = 5.33 Hz, 1H), 4.02–4.07 (m, 1H), 3.14 (dxd, ²*J* = 9.39 Hz, ³*J* = 5.57 Hz, 1H), 3.04 (dxd, ²*J* = 9.39 Hz, ³*J* = 5.57 Hz, 1H), 3.04 (dxd, ²*J* = 9.39 Hz, ³*J* = 5.57 Hz, 1H), 2.42 (d, ³*J* = 6.16, 1H, OH), 1.98 (br s, 3H), 1.72–1.82 (br m, 6H), 1.59–1.68 (br m, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 143.8, 140.9, 128.8, 128.7, 128.1, 127.7, 127.0, 126.7, 86.9, 74.5, 64.9, 48.0, 46.3, 43.9, 36.2, 29.7; HRMS (ES+) *m*/*z* calcd for C₃₈H₄₀O₂NaS: 583.2647 [M + Na]⁺, found 583.2631.

Protection of epoxyalcohol 1b as a Methyl Ether. (2S,3S)-2mesityl-3-(methoxymethyl)oxirane (2f). A solution of 1b (300 mg, 1.56 mmol) in DMF (3 mL) was added via canula to a suspension of sodium hydride (44 mg, 1.81 mmol) in DMF (2 mL) at -20 °C under N₂. The mixture was stirred for 20 min, and methyl iodide (126 µL, 2.03 mmol) was syringed into the mixture. After being stirred for 4 h at -20 °C, the mixture was allowed to reach room temperature and stirred for another hour. MeOH (10 mL) and brine (10 mL) were added. The aqueous solution was extracted with CH_2Cl_2 (3 × 20 mL). The combined organic extracts were dried and concentrated in vacuo. The residual oil was chromatographed using hexane: $Et_2O(9:1-7:3)$ as eluent to give 300 mg (93%) of the epoxyether (2f) as an oil. $[\alpha_D]^{27} - 13.4$ (c 0.44, CHCl₃); IR (neat) 2977, 2887, 1729, 1615, 1453, 1374, 1315, 1199, 954, 850, 816, 703 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.82 (s, 2H), 3.88 (dxd, ²J = 11.4 Hz, ${}^{3}J = 2.9$ Hz, 1H), 3.81 (d, ${}^{3}J = 2.3$ Hz, 1H), 3.56 (dxd, ${}^{2}J =$ $11.4 \text{ Hz}, {}^{3}J = 5.3 \text{ Hz}, 1\text{H}$, 3.46 (s, 3H), 3.13-3.16 (m, 1H), 2.35(s, 6H), 2.26 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 137.4, 137.0, 130.3, 128.6, 72.6, 59.3, 58.3, 54.4, 20.8, 19.6; HRMS (ES+) m/z calcd for $C_{13}H_{18}O_2Na$: 229.1204 $[M + Na]^+$, found 229.1205.

(1*R*,2*S*)-1-(*tert*-Butylthio)-1-mesityl-3-methoxypropan-2-ol (6m). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 85 °C for 20 h: 2f (21 mg, 0.1 mmol), *tert*-butylthiol (98 μL, 0.91 mmol, 9 equiv), and NaOH (37 mg, 0.91 mmol, 9 equiv). The product was purified by flash chromatography (90% hexane,10% ethyl acetate) to yield 6m as a white solid (20 mg, 65% yield): mp 85–86 °C; $[\alpha]^{26}_{D}$ -102.7 (*c* 0.64, CHCl₃); IR (neat) 3444, 2959, 2921, 2896, 2862, 1611, 1456, 1364, 1161, 1120, 1093, 1061, 931, 850, 823, 728 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.83 (s, 1H), 6.79 (s, 1H), 4.42 (d, ³*J* = 10.2 Hz, 1H), 4.04–4.09 (m, 1H), 3.81 (dxd, ²*J* = 9.6 Hz, ³*J* = 2.5 Hz, 1H), 3.73 (dxd, ²*J* = 9.6 Hz, ³*J* = 4.9 Hz, 1H), 3.43 (s, 3H), 2.46 (s, 3H), 2.42 (s, 3H), 2.33 (s, 3H), 1.98 (d, ³*J* = 3.8 Hz, 1H, OH), 1.27 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 137.4, 136.4, 136.2, 134.8, 131.3, 129.2, 73.9, 72.3, 59.1, 44.3, 43.6, 31.4, 21.5, 20.8; HRMS (ES+) *m/z* calcd for C₁₇H₂₈O₂-NaS 319.1708 [M + Na]⁺, found 319.1719.

(1*R*,2*S*)-1-(Cyclohexylthio)-1-mesityl-3-methoxypropan-2-ol (6). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 85 °C for 4 h: 2f (39 mg, 0.19 mmol), cyclohexylthiol (119 μL, 0.95 mmol, 5 equiv), and NaOH (38 mg, 0.95 mmol, 5 equiv). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield **6** as a white solid (50 mg, 82% yield): mp 89–90 °C; $[\alpha]^{27}_{D}$ –146.8 (*c* 0.49, CHCl₃); IR (neat) 3447, 2923, 2850, 1447, 1376, 1262, 1195, 1121, 1094, 1063, 851, 643 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.83 (s, 1H), 6.81 (s, 1H), 4.44 (d, ³*J* = 10.3 Hz, 1H), 4.16–4.22 (m, 1H), 3.82 (dxd, ²*J* = 9.7 Hz, ³*J* = 2.3 Hz, 1H), 3.72 (dxd, ²*J* = 9.7 Hz, ³*J* = 5 Hz, 1H), 2.52–2.59 (m, 1H), 1.94 (d, ³*J* = 3.9 Hz, 1H, OH), 1.55–1.86 (m, 4H), 1.19–1.39 (m, 5H); ¹³C NMR (100 MHz, CDCl₃) δ 139.2, 136.9, 136.4, 133.7, 131.3, 129.1, 74.2, 72.4, 59.1, 45.0, 44.7, 34.2, 34.0, 26.1, 26.0, 25.7, 21.6, 21.1, 20.7; HRMS (ES+) *m*/*z* calcd for C₁₉H₃₀O₂NaS 345.1864 [M + Na]⁺, found 345.1864.

(1R,2S)-1-(Adamanthylthio)-1-mesityl-3-methoxypropan-2-ol (6n). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 90 °C for 24 h: 2f (100 mg, 0.48 mmol), adamanthanethiol (773 mg, 4.4 mmol, 9 equiv), and NaOH (175 mg, 4.4 mmol, 9 equiv). The product was purified by flash chromatography (95% hexane, 5% ethyl acetate) to yield **6n** as a white solid (50 mg, 27% yield):mp 87–88 °C; $[\alpha]^{27}$ _D -149.1 (c 0.58, CHCl₃); IR (neat) 3433, 2902, 2848, 1448, 1230, 1113, 1082, 1071, 1043, 948, 899, 851, 608 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.82 (s, 1H), 6.79 (s, 1H), 4.45 (d, ³J = 10.1 Hz, 1H), 4.02-4.08 (m, 1H), 3.82 (dxd, ${}^{2}J = 9.7$ Hz, ${}^{3}J =$ 2.6 Hz, 1H), 3.69 (dxd, ${}^{2}J = 9.7$ Hz, ${}^{3}J = 5.4$ Hz, 1H), 3.43 (s, 3H), 2.45 (s, 3H), 2.43 (s, 3H), 2.23 (s, 3H), 2 (br s, 4H), 1.61-1.86 (m, 16H); ¹³C NMR (100 MHz, CDCl₃) δ 137.4, 136.2, 136.1, 135.2, 131.2, 129.2, 74.0, 72.3, 59.1, 46.5, 43.8, 40.8,36.2, 29.7, 21.7, 21.52, 20.8; HRMS (ES+) m/z calcd for $C_{23}H_{34}O_2NaS 397.2177 [M + Na]^+$, found 397.2177.

(2S,3R)-3-(tert-Butylthio)-3-mesitylpropane-1,2-diol (8b). The following reagents were combined in the amounts indicated according to the general thiolate ring-opening of epoxides procedure and heated at 80 °C for 3 h: 1b (196 mg, 1.02 mmol), tertbutylthiol (338 µL, 3.06 mmol, 3 equiv), and NaOH (122 mg, 3.06 mmol, 3 equiv). The product was purified by flash chromatography (hexane/ethyl acetate, 7:3-6:4) to yield 8b as a white solid (230 mg, 80% yield): mp 103-104 °C; $[\alpha]^{27}_{D}$ -141.3 (c 0.57, CHCl₃); IR (neat) 3347, 2965, 2920, 2861, 1609, 1456, 1363, 1276, 1216, 1162, 1091, 1030, 1012, 918, 878, 849 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.85 (s, 1H), 6.82 (s, 1H), 4.43 (d, ³J = 10.2 Hz, 1H), 3.97–4.04 (m, 2H), 3.88–3.95 (m, 1H), 2.46 (s, 3H), 2.42 (s, 3H), 2.25 (s, 3H), 2.17–2.22 (br s, 1H, OH), 1.73 (br s, 1H), 1.28 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 137.5, 136.8, 136.7, 134.0, 131.5, 129.4, 73.2, 63.7, 44.9, 31.4, 21.5, 21.5, 20.7; HRMS (ES+) m/z calcd for C₁₆H₂₆O₂NaS 305.1551 [M + Na]⁺, found 305.1563.

(1*R*,2*S*)-1-(*tert*-Butylthio)-1-mesityl-3-(trityloxy)propan-2-ol (9b). A solution of **8b** (282 mg, 1 mmol) and triphenylmethyl chloride (369 mg, 1.3 mmol) in pyridine (7 mL) was heated at 90 °C for 18 h under N₂. The solvent was removed in vacuo, and the residual oil was chromatographed using hexane/Et₂O (9:1–7:1) as eluent to give 350 mg (67%) of **9b** as a white solid: mp 109–110 °C; [α]²⁷_D – 51.7 (*c* 0.58, CHCl₃); IR (neat) 3539, 3057, 3022, 2958, 2921, 1489, 1447, 1363, 1219, 1158, 1091, 1031, 928, 899, 852, 758, 75, 698, 640 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.45–7.5 (m, 5H), 7.22–7.32 (m, 10H), 6.8 (s, 1H), 6.79 (s, 1H), 4.34 (d, ³*J* = 10 Hz, 1H), 3.99–4.04 (m, 1H), 3.72 (dxd, ²*J* = 9.7 Hz, ³*J* = 2.8 Hz, 1H), 3.36 (dxd, ²*J* = 9.7 Hz, ³*J* = 5.5 Hz, 1H), 2.43 (s, 3H), 2.39 (s, 3H), 2.23 (s, 3H), 2.12 (d, ³*J* = 3.8 Hz, 1H, OH), 1.08 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 144.0, 137.5, 136.1, 135.0, 131.2, 129.1, 128.8, 127.8, 127.1, 87.0, 72.6, 65.9, 44.0, 44.0, 31.4, 21.7, 21.5, 20.7; HRMS (ES+) *m*/*z* calcd for C₃₅H₄₀O₂NaS 547.2647 [M + Na]⁺, found 547.2663.

(2R,3R)-3-(Adamantylthio)-2-hydroxy-3-phenylpropanoic acid (12). A mixture of 11 (200 mg, 1.12 mmol), adamanthanethiol (398 mg, 2.24 mmol, 2 equiv), and lithium hydroxide (53.8 mg, 2.24 mmol, 2 equiv) in a mixture of dioxane/H₂O, 10:1 (4.5 mL) was stirred at 80 °C for 1 h. Then 10 mL of water was added, and the mixture was extracted with CH₂Cl₂ to remove the excess thiol. The aqueous phase was acidified with HCl1 M and extracted with AcOEt (3 \times 10 mL). The combined organic phases were dried over MgSO₄ and filtered, and the solvent was removed under reduced pressure. The solid obtained was used without further purification: mp 197–198 °C; [α]²⁶_D –141.27 (*c* 0.52, CHCl₃); IR (neat) 3435, 3350, 3239, 2922, 2846, 1731, 1692, 1448, 1340, 1096, 698 cm^{-1} ; ¹H NMR (400 MHz, CDCl₃) δ 7.51 (m, 2H), 7.32–7.36 (m, 2H), 7.24-7.28 (m, 1H), 4.47 (d, ${}^{3}J = 3.3$ Hz, 1H), 4.42 (d, ${}^{3}J = 3.3$ Hz, 1H), 3.11 (s, 1H, OH), 1.98 (br s, 2H), 1.71-1.83 (br m, 6H), 1.59–1.67 (br m, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 175.0, 141.7, 128.5, 128.3, 127.4, 73.8, 47.9, 46.3, 43.7, 36.1, 29.7; HRMS (ES+) m/z calcd for $C_{19}H_{24}O_3NaS 355.1344 [M + Na]^+$, found 355.1357.

(2R,3R)-3-(Adamantylthio)-3-phenylpropane-1,2-diol (8a-syn). To a solution of 12 (372 mg, 1.12 mmol) in THF (7 mL) at 0 °C was added dropwise the complex borane-dimethyl sulfide (425 μ L, 4.48 mmol, 4 equiv), and the mixture was stirred at rt for 16 h. MeOH was added slowly followed by water. The mixture was extracted with CH_2Cl_2 (2 × 15 mL) and once with 15 mL of EtOAc. The combined organic phases were dried with MgSO₄ and filtered. The solvent was removed in vacuo to yield **8a**-syn as a white solid (302 mg, 84% yield). The solid obtained was used without further purification: mp 95-96 °C; $[\alpha]^{26}_{D}$ –240.6 (c 0.32, CHCl₃); IR (neat) 3499, 3208, 3053, 2898, 2847, 1594, 1108, 1084, 868, 746, 698 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.32–7.2 (m, 5H), 4.07 (d, ³J = 9.6 Hz, 1H), 3.63-3.55 (m, 2H), 3.45 (s. 1H, OH), 3.31-3.26 (m, 1H), 2.1 (br s, 1H, OH), 2.0 (br s, 3H), 1.77-1.88 (br m, 6H), 1.61-1.69 (br m, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 142.3, 128.6, 128.0, 127.1, 73.7, 62.8, 49.4, 47.1, 43.9, 36.1, 29.7; HRMS (ES+) calcd for $C_{19}H_{26}O_2NaS$ 341.1551 [M + Na]⁺, found 341.1548.

(1*R*,2*R*)-1-(Adamantylthio)-1-phenyl-3-(trityloxy)propan-2-ol (9a-syn). A solution of 8a-syn (220 mg, 0.7 mmol) and triphenylmethyl chloride (235 mg, 0.83 mmol) in pyridine (10 mL) was heated at 90 °C for 18 h under N₂. The solvent was removed in vacuo, and the residual oil was chromatographed using hexane/ Et₂O (9:1–7:1) as eluent to give 300 mg (77%) of 9a-syn as a white solid: mp 113–114 °C; [α]²⁸_D –101.84 (*c* 0.91, CHCl₃); IR (neat) 3461, 2903, 2848, 1489, 1448, 1083, 746, 696 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 7.42–7.44 (m, 6H), 7.09–7.27 (m, 14H), 4.3 (d, ³J = 8.5 Hz, 1H), 3.68–3.73 (m, 1H), 3.31 (dxd, ²J = 9.7 Hz, ³J = 3.36 Hz, 1H), 3.16 (d, ³J = 1.75, 1H, OH), 2.72 (dxd, ²J = 9.7 Hz, ³J = 4.1 Hz, 1H), 2.01 (br s, 3H), 1.79–1.9 (br m, 6H), 1.62–1.7 (br m, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 143.99, 142.8, 128.66, 128.2, 128.09, 127.69, 126.87, 126.61, 86.4, 73.77, 63.79, 49.14, 46.75, 44, 36.19, 29.77; HRMS (ES+) *m/z* calcd for C₃₈H₄₀O₂NaS 583.2647 [M + Na]⁺, found 583.2635.

(1R,2S)-1-(Ethoxycarbonyl)-2-hydroxy-2-mesitylethyl 4-Methylbenzenesulfonate (16). To a stirred solution of 15 (2.58 g, 10.22 mmol) in 70 mL of DCM at 0 °C under argon was added triethylamine (2.14 mL, 15.33 mmol, 1.5 equiv) followed by p-toluenesulfonyl chloride (2 g, 10.43 mmol, 1.02 equiv). After being stirred for 70 h at 0 °C, the reaction mixture was diluted with ethyl acetate, and the crude was purified by silica gel chromatography hexane/AcOEt (7:3) to give 16 as a white solid (3.53 g, 85% yield): mp 81-82 °C; $[\alpha]^{26}_{D}$ +17.0 (*c* 0.22, CHCl₃); IR (neat) 3452, 2961, 1704, 1598, 1371, 1310, 1174, 1017, 873, 854 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.78 (d, J = 8.3 Hz, 2H), 7.31 (d, J = 8.3 Hz, 2H), 6.74 (s, 2H), 5.38 (dxd, ³J = 7.9 Hz, ³J = 3.8 Hz, 1H), 5.25 (d, ${}^{3}J = 7.9$ Hz, 1H), 3.74–3.86 (m, 2H), 2.47 (d, ${}^{3}J = 3.8$ Hz, 1H, OH), 2.44 (s, 3H), 2.32 (br s, 6H), 2.2 (s, 3H), 0.84 (t, ${}^{3}J = 7.1$ Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 166.5, 145.2, 138.1, 137.2, 132.7, 130.2, 129.7, 129.1, 126.2, 80.2, 70.9, 61.6, 21.7, 20.8, 20.6, 13.3; HRMS (ES+) m/z calcd for C₂₁H₂₆O₆NaS 429.1348 $[M + Na]^+$, found 429.1337.

(2*S*,3*S*)-Ethyl 3-Mesityloxirane-2-carboxylate (17). A solution of 16 (3.5 g, 8.61 mmol) and 0.78 mL (43.1 mmol, 5 equiv) of water in 45 mL of DMF at rt was treated with potassium carbonate (3.57 g, 25.8 mmol, 3 equiv). After being stirred for 24 h at rt, the reaction mixture was processed with ether in the usual way, and the crude product was purified by silica gel chromatography with 10% ether in hexane to provide 0.51 g (25% yield) of the desired epoxide: $[\alpha]^{27}_{D}$ -13.6 (*c* 0.42, CHCl₃); IR (neat) 2977, 2923, 1755, 1727, 1613, 1442, 1288, 1188, 1107, 1049, 848 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.76 (s, 2H), 4.22 (d, ³J = 4.4 Hz, 1H), 3.87-4.02 (m, 2H), 3.81 (d, ³J = 4.4 Hz, 1H), 2.34 (br s, 6H), 2.23 (s, 3H), 0.91 (t, ³J = 7.1 Hz, 3H); ¹³C

NMR (100 MHz, CDCl₃) δ 167.8, 137.3, 136.8, 128.6, 126.9, 61.0, 57.0, 53.4, 20.9, 19.6, 13.6; HRMS (ES+) *m*/*z* calcd for C₁₄H₁₈O₃Na 257.1154 [M + Na]⁺, found 257.1151.

(2R,3R)-3-(tert-Butylthio)-2-hydroxy-3-mesitylpropanoic acid (18). In a sealed microwave tube were placed the epoxy ester 17 (250 mg, 1.21 mmol), LiOH (87 mg, 3.64 mmol, 3 equiv), and tertbutylthiol (414 μ L, 3.64 mmol) dissolved in 5.5 mL of dioxane/ H_2O (10:1). The mixture was stirred at 140 °C under microwave irradiation for 3 h, 25 mL of water was added, and the mixture was extracted with CH₂Cl₂ to remove the excess thiol. The aqueous phase was acidified with HCl 1 M and was extracted with AcOEt (3 \times 25 mL). The organic phase was dried with MgSO₄ and filtered, and the solvent was removed in vacuo. ¹H NMR spectrum of the resulting crude showed complete conversion and only one regioisomer. The solid obtained (288 mg, 80% yield) was used without further purification:mp 127-128 °C; $[\alpha]^{26}_{D} - 156.13 (c \, 0.25, CHCl_3); IR (neat) 3547, 2961, 1758, 1702, 1456, 1362, 1162, 1101, 929, 852 cm⁻¹; ¹H NMR (400 MHz, CDCl_3) <math>\delta$ 6.83 (s, 1H), 6.8 (s, 1H), 4.62 (d, ³J = 6.8 Hz, 1H), 4.31 $(m, 1H), 3.81 (d, {}^{3}J = 6.8 Hz, 1H), 2.41 (s, 3H), 2.4 (s, 3H), 2.23 (s, 3H), 2.$ 3H), 1.32 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 175.2, 137.3, 136.8, 134.8, 134.5, 131.3, 129.3, 72.2, 47.4, 44.8, 31.4, 21.7, 21.1, 20.8; HRMS (ES-) m/z calcd for C₁₆H₂₃O₃S 295.1368 [M - H]⁻, found 295.1374.

(2R,3R)-3-(tert-Butylthio)-3-mesitylpropane-1,2-diol (8b-syn). To a solution of 18 (504 mg, 1.7 mmol) in THF (10 mL) at 0 °C was added dropwise 646 µL (44.8 mmol, 4 equiv) of the complex borane-dimethyl sulfide. The reaction was stirred at rt for 24 h. Then MeOH was added slowly followed by water. The mixture was extracted with CH_2Cl_2 (2 × 20 mL) and once with 25 mL of EtOAc. The combined organic phases were dried with MgSO4 and filtered, and the solvent was removed under reduced pressure. The crude was purified by silica gel chromatography (hexane/EtOAc) (3:1) to afford 8b-syn (410 mg, 85% yield) as an oil: [α]²⁶_D -192.9 (c 0.67, CHCl₃); IR (neat) 3398, 2959, 2920, 1612, 1457, 1365, 1270, 1160, 1093, 1039, 920, 875, 852, 817, 703 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.81 (s, 2H), 4.4 (d, ${}^{3}J = 10.2$ Hz, 1H), 3.77–3.82 (m, 2H), 3.62–3.66 (m, 2H), 3.25-3.31 (m, 1H), 2.46 (s, 3H), 2.34 (s, 3H), 2.23 (s, 3H), 2.12-2.15 (m, 1H), 1.35 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 136.9, 136.4, 135.4, 134.7, 131.1, 129.3, 71.4, 62.7, 47.0, 44.8, 31.6, 21.6, 21.2, 20.7; HRMS (ES+) m/z calcd for C₁₆H₂₆O₂NaS $305.1551 [M + Na]^+$, found 305.1541.

(1*R*,2*R*)-1-(*tert*-Butylthio)-1-mesityl-3-(trityloxy)propan-2-ol (9b-syn). A solution of 8b-syn (430 mg, 1.52 mmol) and triphenylmethyl chloride (519 mg, 1.83 mmol) in pyridine (15 mL) was heated at 90 °C for 24 h under N₂. The solvent was removed in vacuo, and the residual oil was chromatographed using hexane/ Et₂O (9:1–7:1) as eluent to give 530 mg (66%) of 9b-syn as a white solid: mp 117–118 °C; $[\alpha]^{27}_{D}$ –110.4 (*c* 1.01, CHCl₃); IR (neat) 3483, 3084, 2959, 1596, 1407, 1112, 929, 852 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.33–7.36 (m, 6H), 7.15–7.23 (m, 9H), 6.71 (s, 1H), 6.6 (s, 1H), 4.27 (d, ³J = 9.6 Hz, 1H), 3.95–4 (m, 1H), 3.48 (s, 1H), 3.11 (dxd, ²J = 9.8 Hz, ³J = 3.4 Hz, 1H), 2.92 (dxd, ²J = 9.8 Hz, ³J = 5.1 Hz, 1H), 2.28 (s, 3H), 2.18 (s, 3H), 2.13 (s, 3H), 1.34 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 143.8, 136.7, 135.9, 135.2, 134.7, 131.0, 129.0, 128.7, 127.6, 126.8, 86.9, 71.6, 65.5, 48.1, 44.4, 31.7, 21.7, 21.4, 20.7; HRMS (ES+) *m/z* calcd for C₃₅H₄₀O₂NaS 547.2647 [M + Na]⁺, found 547.2671.

General Procedure for the Phosphinite Incorporation. To a Schlenk flask containing β -hydroxy sulfide (0.55 mmol) and DMAP (0.055 mmol, 0.1 equiv) in toluene (0.28 M) at rt were added NEt₃ (0.66 mmol, 1.2 equiv) and chlorodiphenylphosphine (0.55 mmol, 1.01 equiv) via syringe. The reaction was stirred for 20 min. The solvent was removed in vacuo, and the reaction crude was diluted with 95:5 hexane–ethyl acetate mixture (0.5 mL, dried over molecular sieves, and degassed with argon). The resulting slurry was loaded onto a plug of silica and

purified by flash chromatography (1 \times 3 cm, 95% hexane, 5% ethyl acetate).

((1R,2S)-3-Methoxy-1-phenyl-1-(phenylthio)propan-2-yloxy)diphenylphosphine (4a). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3a (151 mg, 0.55 mmol), DMAP (6.7 mg. 0.055 mmol), NEt₃ (92 µL, 0.66 mmol), and chlorodiphenylphosphine (101.1 μ L, 0.55 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1 × 3 cm, 95% hexane, 5% ethyl acetate) to yield **4a** as an oil (230 mg, 91% yield): $[\alpha]^{26}{}_{\rm D}$ –130.0 (*c* 0.63, CHCl₃); IR (neat) 3067, 2890, 1582, 1492, 1480, 1434, 1127, 1089, 1069, 1024, 998, 965, 953, 739, 691 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.09–7.37 (m, 20H), 4.41–4.46 (m, 2H), 3.37 (dxd, ²J = 9.9 Hz, ${}^{3}J = 4.44$ Hz, 1H), 3.28 (dxd, ${}^{2}J = 9.9$ Hz, ${}^{3}J = 4.78$ Hz, 1H), 2.98 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 143.1 (d, J = 18.9 Hz), 142.1 (d, J = 14.5 Hz), 138.6, 134.9, 132.2, 130.6, 130.4, 130.4, 130.1, 129.3, 129.0, 128.9, 128.7, 128.1, 128.0, 128.0, 127.9, 127.3, 127.1, 81.5 (d, J = 19.7 Hz), 73.2 (d, J = 3,3 Hz), 58.5, 55.6 (d, J = 6.6 Hz); ³¹P NMR δ 119.6; HRMS (ES+) m/z calcd for C₂₈H₂₇O₂NaPS 481.1367 [M + Na]⁺, found 481.1386.

((1R,2S)-3-(Benzyloxy)-1-phenyl-1-(phenylthio)propan-2-yloxy)diphenylphosphine (4b). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3b (36 mg, 0.1 mmol), DMAP (1.3 mg, 0.01 mmol), NEt₃ (17.2 µL, 0.12 mmol), and chlorodiphenylphosphine (19 μ L, 0.1 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1×3 cm, 95%hexane, 5% ethyl acetate) to yield **4b** as an oil (45 mg, 82% yield): $[\alpha]_{D}^{26}$ –117.9 (c 0.51, CHCl₃); IR (neat) 3055, 3027, 3002, 2898, 2857, 1582, 1493, 1479, 1434, 1116, 1092, 1070, 1051, 1024, 998, 736, 691 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.39–7.43 (m, 2H), $7.13-7.33 \text{ (m, 23H)}, 4.53-4.6 \text{ (m, 2H)}, 4.25 \text{ (d, }^{2}J = 11.9 \text{ Hz}, 1\text{H)},$ $4.21 (d, {}^{2}J = 11.9 Hz, 1H), 3.58 - 3.62 (m, 1H), 3.45 - 3.49 (m, 1H);$ ¹³C NMR (100 MHz, CDCl₃) δ 143.0 (d, J = 18.4 Hz), 142.1 (d, J = 14.7 Hz), 138.6, 137.9, 135.0, 132.1, 130.7, 130.5, 130.4, 129.4, 129.0, 128.9, 128.7, 128.2, 128.1, 128.1, 128.0, 128.0, 127.9, 127.7, 127.5, 127.3, 127.0, 81.6 (d, *J* = 19.7 Hz), 73.1, 71.0 (d, *J* = 3,6 Hz), 55.7; ³¹P NMR δ 119.1. HRMS (ES+) m/z calcd for C₃₄H₃₂O₂PS $535.1861 [M + H]^+$, found 535.1853.

((1*R*,2*S*)-3-(Benzhydryloxy)-1-phenyl-1-(phenylthio)propan-2yloxy)diphenylphosphine (4c): $[α]^{27}{}_D$ -59.13 (*c* 0.44, CHCl₃); IR (neat) 3055, 3027, 3003, 2916, 2857, 1583, 1492, 1479, 1435, 1091, 1071, 1047, 1025, 1000, 923, 740, 693 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.26-7.30 (m, 2H), 7.06-7.19 (m, 28H), 5.02 (s, 1H), 4.46-4.55 (m, 2H), 3.54 (dxd, ²*J* = 10.2 Hz, ³*J* = 4.8 Hz, 1H), 3.40 (dxd, ²*J* = 10.2 Hz, ³*J* = 5.16 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 142.8 (d, *J* = 18.3 Hz), 142 (d, *J* = 15.23 Hz), 141.94, 141.91, 138.6, 134.99, 132.03, 130.68, 130.47, 130.38, 130.16, 129.4, 129.97, 128.93, 128.72, 128.25, 128.2, 128.08, 128.03, 128.0, 127.98, 127.91, 127.34, 127.27, 127.21, 126.98, 126.87, 84.15, 81.73 (d, *J* = 19.81 Hz), 69.63 (d, *J* = 3,78 Hz), 58.56, 57.72 (d, *J* = 6.03 Hz); ³¹P NMR (162, MHz, CDCl₃) δ 118.38; HRMS (ES+) *m*/*z* calcd for C₄₀H₃₅O₂NaSP: 633.1993 [M + Na]⁺, found 633.2010.

Diphenyl((1*R*,2*S*)-1-phenyl-1-(phenylthio)-3-(trityloxy)propan-2-yloxy)phosphine (4d). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3d (85.6 mg, 0.17 mmol), DMAP (2.1 mg, 0.017 mmol), NEt₃ (29 μ L, 0.2 mmol), and chlorodiphenylphosphine (31.5 μ L, 0.17 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1 × 3 cm, 95% hexane, 5% ethyl acetate) to yield 4d as an oil (103 mg, 88% yield): [α]²⁵_D -56.7 (*c* 0.67, CHCl₃); IR (neat) 3056, 3025, 3004, 2922, 2872, 2852, 1583, 1491, 1480, 1448, 1436, 1218, 1072, 1026, 1000, 747, 696 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.01–7.30 (m, 35H), 4.50 (d, ³J = 4.98 Hz, 1H), 4.39–4.45 (m, 1H), 3.14 (dxd, ${}^{2}J = 9.8$ Hz, ${}^{3}J = 5.7$ Hz, 1H), 3.06 (dxd, ${}^{2}J = 9.8$ Hz, ${}^{3}J = 5.8$ Hz, 1H); 13 C NMR (100 MHz, CDCl₃) δ 143.7, 142.8 (d, J = 18.5 Hz), 142.1 (d, J = 15.5 Hz), 138.0, 135.5, 131.6, 130.8, 130.6, 130.4, 130.2, 129.4, 129.0, 128.9, 128.7, 128.6, 128.0, 128.0, 127.9, 127.7, 127.2, 126.9, 126.8, 87.1, 82.1 (d, J = 19.4 Hz), 64.6 (d, J = 4.2 Hz), 56.1 (d, J = 5.2 Hz); 31 P NMR (162, MHz, CDCl₃) δ 118.8; HRMS (ES+) m/z calcd for C₄₆H₃₉O₂NaSP 709.2306 [M + Na]⁺, found 709.2307.

((1R,2S)-1-(3,5-Dimethylphenylthio)-3-methoxy-1-phenylpropan-2-yloxy)diphenylphosphine (4e). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3e (81.7 mg, 0.27 mmol), DMAP (3.3 mg, 0.027 mmol), NEt₃ (45 µL, 0.32 mmol), and chlorodiphenylphosphine (50 μ L, 0.27 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1 \times 3 cm, 95% hexane, 5% ethyl acetate) to yield **3a** as an oil (115 mg, 87% yield): $[\alpha]^{27}_{D}$ -119.8 (c 0.66, CHCl₃); IR (neat) 3052, 3027, 2916, 2893, 1599, 1580, 1492, 1480, 1466, 1453, 1434, 1129, 1094, 1074, 1026, 967, 848, 741 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.41-7.46 (m, 2H), 7.36-7.38 (m, 2H), 7.21-7.33 (m, 11H), 6.91 (s, 2H), 6.81 (s, 1H), 4.47-4.54 (m, 2H), 3.46 (dxd, ${}^{2}J = 9.9$ Hz, ${}^{3}J = 4.4$ Hz, 1H), 3.38 (dxd, ${}^{2}J =$ 9.9 Hz, ${}^{3}J = 4.4$ Hz, 1H), 3.06 (s, 3H), 2.21 (s, 6H); ${}^{13}C$ NMR $(100 \text{ MHz}, \text{CDCl}_3) \delta 143.3 (d, J = 18.5 \text{ Hz}), 142.2 (d, J = 14.6 \text{ Hz}),$ 138.8, 138.2, 134.3, 130.5 (d, J = 22.0 Hz), 130.2 (d, J = 22.0 Hz), 129.7, 129.3, 129.0, 128.9, 128.8, 128.1, 128.0, 127.9 (d, J = 7.3 Hz),127.2, 81.5 (d, J = 19.7 Hz), 73.2 (d, J = 3.7 Hz), 58.5, 55.3 (d, J = 6.5 Hz), 21.1; ³¹P NMR (162, MHz, CDCl₃) δ 119.3; HRMS (ES+) m/z calcd for C₃₀H₃₁O₂NaPS 509.1680 [M + Na]⁺, found 509.1688

((1R,2S)-1-(2,6-Dimethylphenylthio)-3-methoxy-1-phenylpropan-2-yloxy)diphenylphosphine (4f). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3f (87.6 mg, 0.27 mmol), DMAP (3.3 mg, 0.027 mmol), NEt₃ (45 µL, 0.32 mmol), and chlorodiphenylphosphine ($50 \,\mu$ L, 0.27 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1×3 cm, 95% hexane, 5% ethyl acetate) to yield **4f** as an oil (110 mg, 80% yield): $[\alpha]^{28}{}_{\rm D}$ -174.4 (*c* 0.62, CHCl₃); IR (neat) 3052, 2921, 2891, 2859, 2807, 1478, 1457, 1433, 1124, 1106, 1095, 1076, 1060, 1009, 947, 876, 777, 736, 692 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.41-7.45 (m, 2H), 7.28-7.32 (m, 3H), 7.11-7.20 (m, 10H), 7.02-7.06 (m, 2H), 6.97-6.99 (m, 2H), 4.59-4.66 (m, 1H), 4.15 (d, ${}^{3}J = 6.7$ Hz, 1H), 3.49 (dxd, ${}^{2}J = 10$ Hz, ${}^{3}J = 4.98$ Hz, 1H), 3.43 (dxd, ${}^{2}J =$ 10 Hz, ${}^{3}J = 4.4$ Hz, 1H), 3.03 (s, 3H), 2.28 (s, 6H); ${}^{13}C$ NMR $(100 \text{ MHz}, \text{CDCl}_3) \delta 143.6, 143.2 \text{ (d}, J = 19 \text{ Hz}), 142.1 \text{ (d}, J =$ 14 Hz), 139.5, 132.6, 130.4 (d, J = 19.1 Hz), 130.2 (d, J =18 Hz), 129.0, 129.0, 128.6, 128.3, 128.1, 128.0 (d, J = 7.3 Hz), 128.0, 127.8 (d, J = 7 Hz), 127.0, 82.4 (d, J = 19.7 Hz), 73.6 (d, J = 2.93 Hz), 58.4, 55.4 (d, J = 6.56 Hz), 21.9; ³¹P NMR (162, MHz, CDCl₃) δ 120.4; HRMS (ES+) m/z calcd for C₃₀H₃₁O₂NaPS 509.1680 $[M + Na]^+$, found 509.1683.

((1*R*,2*S*)-3-Methoxy-1-(4-methoxyphenylthio)-1-phenylpropan-2-yloxy)diphenylphosphine (4g). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3g (82.2 mg, 0.27 mmol), DMAP (3.3 mg, 0.027 mmol), NEt₃ (45 μ L, 0.32 mmol), and chlorodiphenylphosphine (50 μ L, 0.27 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1 × 3 cm, 95% hexane, 5% ethyl acetate) to yield 3a as an oil (112 mg, 85% yield): [α]²⁷_D –93.7 (*c* 0.65, CHCl₃); IR (neat) 2924, 2893, 2835, 1591, 1571, 1493, 1454, 1438, 1285, 1246, 1181, 1126, 1104, 1072, 720, 640 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.33–7.37 (m, 2H), 7.12–7.22 (m, 15H), 6.64–6.66 (m, 2H), 4.38–4.44 (m, 1H), 4.26 (d, ³J = 5.85 Hz, 1H), 3.68 (s, 3H), 3.38 (dxd, ²J = 9.8 Hz, ³J = 5.26 Hz, 1H), 3.31 (dxd, ²J = 9.8 Hz, ³J = 4.82 Hz, 1H), 2.97 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 159.6, 143.3 (d, J = 19 Hz), 142.3 (d, J = 14 Hz), 138.8, 135.6, 130.5, 130.4, 130.3, 130.2, 129.4, 129.0, 128.8, 128.1, 128.0, 128.0, 127.9, 127.9, 127.1, 124.9, 114.3, 81.2 (d, J = 19.3 Hz), 73.4 (d, J = 3 Hz), 58.5, 56.9 (d, J = 6.56 Hz), 52.2; ³¹P NMR (162, MHz, CDCl₃) δ 119.4; HRMS (ES+) m/z calcd for C₂₉H₃₀O₃PS 489.1653 [M + H]⁺, found 489.1652.

((1R,2S)-1-(4-Bromophenylthio)-3-methoxy-1-phenylpropan-2-yloxy)diphenylphosphine (4h). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: **3h** (95.4 mg, 0.27 mmol), DMAP (3.3 mg, 0.027 mmol), NEt₃ (45 µL, 0.32 mmol), and chlorodiphenylphosphine (50 μ L, 0.27 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1×3 cm, 95% hexane, 5% ethyl acetate) to yield **4h** as an oil (129 mg, 89% yield): $[\alpha]^{27}{}_{\rm D} - 100.6$ (c 0.31, CHCl₃); IR (neat) 3053, 2891, 1492, 1472, 1453, 1385, 1231, 1190, 1129, 1091, 1070, 1026, 1009, 887, 741, 698 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.32-7.36 (m, 2H), 7.14-7.27 (m, 15H), 7.01–7.05 (m, 2H), 4.34–4.44 (m, 2H), 3.36 (dxd, ${}^{2}J =$ 9.8 Hz, ${}^{3}J = 4.5$ Hz, 1H), 3.23 (dxd, ${}^{2}J = 9.8$ Hz, ${}^{3}J = 5.11$ Hz, 1H), 3.01 (s, 3H); ${}^{13}C$ NMR (100 MHz, CDCl₃) δ 143.0 (d, J =18.9 Hz), 142.0 (d, J = 15.5 Hz), 138.1, 134.1, 133.6, 131.8, 130.7, 130.5, 130.3, 130.1, 129.4, 129.1, 129.0, 128.2, 128.1, 128.1, 128.1, 128.0, 127.5, 121.3, 81.2 (d, J = 19.7 Hz), 73.0 (d, J = 3,6 Hz), 58.6, 55.7 (d, J = 5.9 Hz); ³¹P NMR (162, MHz, CDCl₃) δ 119.4; HRMS (ES+) m/z calcd for C₂₈H₂₇O₂PBrS $537.0653 [M + H]^+$, found 537.0634.

((1R,2S)-1-(4-tert-Butylphenylthio)-3-methoxy-1-phenylpropan-2-yloxy)diphenylphosphine (4i). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3i (86 mg, 0.26 mmol), DMAP (3.2 mg, 0.026 mmol), NEt₃ (44 µL, 0.31 mmol), and chlorodiphenylphosphine (48 µL, 0.26 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography $(1 \times 3 \text{ cm}, 95\% \text{ hexane}, 5\% \text{ ethyl acetate})$ to yield **4i** as an oil (107 mg, 80% yield): $[\alpha]_{D}^{26} = -103.6$ (c 0.61, CHCl₃); IR (neat) 3053, 2959, 1491, 1481, 1452, 1434, 1394, 1362, 1190, 1128, 1120, 1073, 741, 697 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.21-7.44 (m, 19H), 4.45–4.53 (m, 2H), 3.42 (dxd, ${}^{2}J = 9.9$ Hz, ${}^{3}J = 5.3$ Hz, 1H), 3.35 (dxd, ${}^{2}J = 9.9$ Hz, ${}^{3}J = 4.84$ Hz, 1H), 3.02 (s, 3H), 1.26 (s, 9H); ${}^{13}C$ NMR (100 MHz, CDCl₃) δ 150.4, 143.2 (d, J = 18.3 Hz), 142.2 (d, J = 14.3 Hz), 138.9, 132.2, 131.4, 130.6, 130.4, 130.4, 130.2, 129.3, 129.0, 128.5, 128.1, 128.1, 128.0, 128.0, 127.9, 127.2, 125.8, 81.5 (d, J = 19.6 Hz), 73.3 (d, J = 3.1 Hz), 58.5, 55.8 (d, J = 6.7 Hz), 34.5, 31.2; ³¹P NMR (162, MHz, CDCl₃) δ 119.3; HRMS (ES+) m/z calcd for $C_{32}H_{35}O_2NaSP$ 537.1993 [M + Na]⁺, found 537.2002.

((1R,2S)-3-Methoxy-1-(naphthalen-2-ylthio)-1-phenylpropan-2-yloxy)diphenylphosphine (4j). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3j (87.6 mg, 0.27 mmol), DMAP (3.3 mg, 0.027 mmol), NEt₃ (45 µL, 0.32 mmol), and chlorodiphenylphosphine ($50 \,\mu$ L, 0.27 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1 \times 3 cm, 95% hexane, 5% ethyl acetate) to yield **4j** as an oil (110 mg, 80% yield): $[\alpha]_{D}^{27}$ -52.2 (*c* 0.42, CHCl₃); IR (neat) 3053, 2922, 2890, 2858, 1588, 1493, 1453, 1437, 1227, 1194, 1125, 1114, 965, 943, 744, 728, 695 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.74-7.78 (m, 2H), 7.65-7.69 (m, 2H), 7.20–7.47 (m, 18H), 4.68 (d, ${}^{3}J = 5.9$ Hz, 1H), 4.54–4.60 (m, 1H), 3.50 (dxd, ${}^{2}J = 9.9$ Hz, ${}^{3}J = 5.5$ Hz, 1H), 3.4 (dxd, $^{2}J = 9.9$ Hz, $^{3}J = 5.3$ Hz, 1H), 3.07 (s, 3H); 13 C NMR (100 MHz, $CDCl_3$) δ 143.1 (d, J = 18.3 Hz), 142.1 (d, J = 14.6 Hz), 138.5, 133.6, 132.4, 132.2, 130.7, 130.5, 130.3, 130.1, 129.4, 129.4, 129.0, 129.0, 128.2, 128.2, 128.1, 128.0, 128.0, 127.9, 127.6, 127.4, 127.4, 126.3, 126.0, 81.4 (d, J = 19.76 Hz), 73.1 (d, J = 3.7 Hz), 58.6, 55.4 (d, J = 6.2 Hz); ³¹P NMR (162, MHz, CDCl₃) δ 119.6; HRMS (ES+) m/z calcd for C₃₂H₂₉O₂NaPS 531.1524 $[M + Na]^+$, found 531.1523.

((1R,2S)-1-(Isopropylthio)-3-methoxy-1-phenylpropan-2-yloxy)diphenylphosphine (4k). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3k (64.9 mg, 0.27 mmol), DMAP (3.3 mg, 0.027 mmol), NEt₃ (45 µL, 0.32 mmol), and chlorodiphenylphosphine (50 μ L, 0.27 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography $(1 \times 3 \text{ cm}, 95\% \text{ hexane}, 5\% \text{ ethyl acetate})$ to yield **4k** as an oil $(99 \text{ mg}, 86\% \text{ yield}): [\alpha]^{28} - 106.9 (c \, 0.58, \text{CHCl}_3); \text{IR (neat) 3160}, 3139, 3125, 2957, 1738, 1492, 1480, 1453, 1434, 1381, 1365, 1230, 1230, 1434, 1434, 1381, 1365, 1230, 1434, 1381, 1$ 1216, 1200, 1130, 1095, 1073, 1054, 1026, 760, 741, 699 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.40–7.48 (m, 4H), 7.16–7.34 (m, 11H), 4.41–4.48 (m, 1H), 4.24 (d, ${}^{3}J$ = 6.7 Hz, 1H), 3.55 (dxd, $^{2}J = 10.23$ Hz, $^{3}J = 4.97$ Hz, 1H), 3.47 (dxd, $^{2}J = 10.23$ Hz, $^{3}J = 1$ 4.38 Hz, 1H), 3.17 (s, 3H), 2.63–2.73 (sept, ${}^{3}J = 6.7$ Hz, 1H), 1.24 (d, ${}^{3}J = 6.7$ Hz, 3H), 1.16 (d, ${}^{3}J = 6.7$ Hz, 3H); ${}^{13}C$ NMR $(100 \text{ MHz}, \text{CDCl}_3) \delta 143.2 \text{ (d}, J = 19 \text{ Hz}), 142.1 \text{ (d}, J = 14 \text{ Hz}),$ 140.0, 130.5, 130.4, 130.3, 130.2, 129.2, 128.9, 128.7, 128.1, 128.0, 128.0, 127.8, 127.8, 127.0, 82.4 (d, J = 19 Hz), 73.3 (d, J = 3,42 Hz), 58.6, 50.6 (d, J = 5.8 Hz), 34.7, 23.4, 23.2; ³¹P NMR (162, MHz, CDCl₃) δ 119.2; HRMS (ES+) m/z calcd for C₂₅H₂₉O₂P-NaS 447.1524 $[M + Na]^+$, found 447.1517.

((1R,2S)-1-(Cyclohexylthio)-3-methoxy-1-phenylpropan-2-yloxy)diphenylphosphine (41). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 31 (31 mg, 0.11 mmol), DMAP (1.4 mg, 0.011 mmol), NEt₃ (19 µL, 0.13 mmol), and chlorodiphenylphosphine (20 μ L, 0.11 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography $(1 \times 3 \text{ cm}, 95\% \text{ hexane}, 5\% \text{ ethyl acetate})$ to yield **4** as an oil $(35 \text{ mg}, 68\% \text{ yield}): [\alpha]^{28}_{D} - 104.9 (c \, 0.54, \text{CHCl}_3); \text{IR (neat) 2924},$ 2850, 1492, 1480, 1449, 1435, 1129, 1096, 1070, 927, 885, 741, 697 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.14–7.46 (m, 15H), 4.38–4.44 (m, 1H), 4.23 (d, ${}^{3}J = 6.72$ Hz, 1H), 3.54 (dxd, ${}^{2}J = 10$ Hz, ${}^{3}J = 6.72$ Hz, 1H), 3.44 (dxd, ${}^{2}J = 10$ Hz, ${}^{3}J = 4.4$ Hz, 1H), 3.14 (s, 3H), 2.42-2.50 (m, 1H), 1.92-1.95 (br d, 1H), 1.63-1.79 (m, 4H), 1.14-1.32 (m, 5H); 13 C NMR (100 MHz, $CDCl_3$) δ 143.2 (d, J = 18.3 Hz), 142.1 (d, J = 14.54 Hz), 140.2, 130.5, 130.4, 130.3, 130.2, 129.1, 129.0, 128.7, 128.2, 128.0, 128.0, 127.9, 127.8, 127.0, 82.6 (d, J = 19 Hz), 73.4 (d, J = 3.3 Hz), 58.6, 50.0 (d, J = 6.6 Hz), 43.3, 33.5, 29.7, 26.0, 25.8, 25.8; ³¹P NMR (162, MHz, CDCl₃) δ 119.0; HRMS (ES+) m/z calcd for $C_{28}H_{33}O_2NaPS$ 487.1837 [M + Na]⁺, found 487.1830.

((1R,2S)-1-(tert-Butylthio)-3-methoxy-1-phenylpropan-2-yloxy)diphenylphosphine (4m). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3m (53.4 mg, 0.21 mmol), DMAP (2.6 mg, 0.021 mmol), NEt₃ (35 µL, 0.25 mmol), and chlorodiphenylphosphine (39 μ L, 0.21 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography $(1 \times 3 \text{ cm}, 95\% \text{ hexane}, 5\% \text{ ethyl acetate})$ to yield **4m** as an oil $(84 \text{ mg}, 91\% \text{ yield}): [\alpha]^{27}_{D} - 116.6 (c \ 0.63, \text{CHCl}_3); \text{ IR (neat) } 3056,$ 3000, 2960, 2938, 2895, 1492, 1480, 1471, 1453, 1437, 1364, 1189, 1127, 699 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.16–7.44 (m, 15H), 4.32–4.39 (m, 1H), 4.19 (d, ${}^{3}J = 6.4$ Hz, 1H), 3.49 (dxd, ${}^{2}J = 9.7$ Hz, ${}^{3}J = 4.97$ Hz, 1H), 3.32 (dxd, ${}^{2}J = 9.7$ Hz, ${}^{3}J$ 5.1 Hz, 1H), 3.13 (s, 3H), 1.21 (s, 9H); ¹³C NMR (100 MHz, $CDCl_3$) δ 143.3 (d, J = 19 Hz), 142.1 (d, J = 14 Hz), 141.9, 130.5 (d, J = 13.8 Hz), 130.355 (d, J = 15 Hz), 129.452, 129.0, 128.7,128.0, 128.0, 127.8 (d, J = 6.8 Hz), 126.7, 83.3 (d, J = 19 Hz), 73.0 (d, J = 3.75 Hz), 58.5, 49.2 (d, J = 5.8 Hz), 44.2, 31.3; ³¹P NMR (162, MHz, CDCl₃) δ 118.7; HRMS (ES+) m/z calcd for C₂₆H₃₁- $O_2PNaS 461.1680 [M + Na]^+$, found 461.1682.

 $4m \cdot PdCl_2$. To a Schlenk flask containing the thiophosphinite ligand 4m (2.02 mg, 0.0046 mmol) in CH₂Cl₂ (0.1 mL) was added (MeCN)₂PdCl₂ (1.19 mg 0.0046 mmol), and the reaction was stirred for 30 min. The solution was concentrated in vacuo, and the residue diluted with CH₂Cl₂ (0.1 mL). Et₂O (1.1 mL)

was then added rapidly with stirring to precipitate $4\mathbf{m} \cdot \mathbf{PdCl}_2$, which was filtered and dried in vacuo overnight. An X-ray quality crystal was obtained by slow diffusion of Et_2O into a solution of $4\mathbf{m} \cdot \mathbf{PdCl}_2$ in CH_2Cl_2 .

((1R,2S)-1-(Adamantylthio)-3-methoxy-1-phenylpropan-2-yloxy)diphenylphosphine (4n). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 3n (37 mg, 0.11 mmol), DMAP (1.3 mg, 0.011 mmol), NEt₃ (19 µL, 0.13 mmol), and chlorodiphenylphosphine (20 μ L, 0.11 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography $(1 \times 3 \text{ cm}, 95\% \text{ hexane}, 5\% \text{ ethyl acetate})$ to yield **4n** as an oil $(45 \text{ mg}, 78\% \text{ yield}): [\alpha]^{28}_{D} - 129.7 (c \, 0.49, \text{CHCl}_3); \text{IR (neat) } 2901,$ 2847, 1492, 1480, 1450, 1434, 1343, 1300, 1128, 1095, 1071, 966, 945, 924, 751, 694 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.36–7.44 (m, 4H), 7.14–7.31 (m, 14H), 4.30–4.36 (m, 1H), 4.26 (d, ${}^{3}J = 6.31$ Hz, 1H), 3.50 (dxd, ${}^{2}J = 9.8$ Hz, ${}^{3}J = 5.01$ Hz, 1H), 3.33 (dxd, ${}^{2}J = 9.8$ Hz, ${}^{3}J = 5.02$ Hz, 1H), 3.12 (s, 3H), 1.94 (br s, 3H), 1.71–1.81 (br m, 6H), 1.57–1.64 (br m, 6H); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3) \delta 143.3 \text{ (d}, J = 19.3 \text{ Hz}), 142.4, 142.2 \text{ (d}, J =$ 13.9 Hz), 130.6, 130.5, 130.4, 130.2, 129.3, 129.0, 128.7, 128.0, 128.0, 128.0, 127.8, 126.6, 83.6 (d, J = 19 Hz), 73.1 (d, J =3,3 Hz), 58.5, 46.5 (d, J = 6.5 Hz), $43.7, 43.7, 36.2, 29.7; {}^{31}$ P NMR (162, MHz, CDCl₃) δ 118.7; HRMS (ES+) m/z calcd for C₃₂H₃₈O₂PS: 517.2330 [M + H]⁺, found 517.2302.

((1R,2S)-1-(Adamantylthio)-1-phenyl-3-(trityloxy)propan-2yloxy)diphenylphosphine (10a). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 9a (100 mg, 0.18 mmol), DMAP (2.2 mg, 0.018 mmol), NEt₃ (30 µL, 0.21 mmol), and chlorodiphenylphosphine (33 μ L, 0.18 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography $(1 \times 3 \text{ cm}, 95\% \text{ hexane},$ 5% ethyl acetate) to yield 10a as an oil (120 mg, 90% yield): $[\alpha]^{26}_{D}$ -88.5 (c 0.39, CHCl₃); IR (neat) 3054, 2901, 2847, 1738, 1490, 1448, 1434, 1350, 1299, 1217, 1155, 1071, 1029, 977, 924, 898, 849, 742, 694 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.06–7.43 (m, 30H), 4.37–4.43 (m, 1H), 4.35 (d, ${}^{3}J = 4.31$ Hz, 1H), 3.20 (dxd, ${}^{2}J = 9.8$ Hz, ${}^{3}J = 5.7$ Hz, 1H), 2.96 (dxd, ${}^{2}J = 9.8 \text{ Hz}, {}^{3}J = 6.3 \text{ Hz}, 1\text{H}, 1.93 \text{ (br s}, 3\text{H}), 1.56 - 1.78 \text{ (br m},$ 12H); ¹³C NMR (100 MHz, CDCl₃) δ 143.8, 143.0 (d, J = 19.8 Hz), 142.2 (d, J = 14.61 Hz), 141.1, 130.7 (d, J = 21.3 Hz), 130.5 (d, J = 22.0 Hz), 129.5, 128.9, 128.8, 128.6, 128.0 (d, J = 6.0 Hz), 127.9 (d, J = 6.0 Hz), 127.7, 127.7, 126.9, 126.4, 87.0, 84.4 (d, J = 19 Hz), 64.6 (d, J = 3.8 Hz), 46.9 (d, J = 5.7 Hz), 46.3, 43.7, 36.2, 29.7; ³¹P NMR (162, MHz, CDCl₃) δ 117.7; HRMS (ES+) m/z calcd for C₅₀H₅₀O₂PS 745.3269 [M + H]⁺, found 745.3264.

((1R,2S)-1-(Cyclohexylthio)-1-mesityl-3-methoxypropan-2yloxy)diphenylphosphine (71). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 61 (38 mg, 0.1 mmol), DMAP (1.2 mg, 0.01 mmol), NEt₃ (17 μ L, 0.12 mmol), and chlorodiphenylphosphine (19 µL, 0.1 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1 \times 3 cm, 95% hexane, 5% ethyl acetate) to yield 71 as an oil (41 mg, 79% yield): $[\alpha]^{27}_{D}$ -41.0 (*c* 0.36, CHCl₃); IR (neat) 2924, 2851, 1738, 1435, 1375, 1217, 1094, 1052, 946, 738, 697 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.38-7.42 (m, 1H), 7.23-7.30 (m, 3H), 7.1-7.13 (m, 1H), 6.97-7.01 (m, 2H), 6.64-6.67 (m, 4H), 4.62 (d, ${}^{3}J = 10.9$ Hz, 1H), 4.48-4.54 (m, 1H), 3.82 (s, 1H), 3.81 (s, 1H), 3.18 (s, 3H), 2.57-2.64 (m, 1H), 2.38 (s, 3H), 2.29 (s, 3H), 2.20 (s, 3H), 2-2.04 (m, 1H), 1.86-1.91 (m, 1H), 1.66–1.78 (m, 2H), 1.52–1.58 (m, 1H), 1.17–1.37 (m, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 143.7, (d, J = 17.6 Hz), 142.2 (d, J = 14.7 Hz), 136.8, 136.5, 135.7, 134.9, 131.0, 130.0(d, J = 21.6 Hz), 130.0 (d, J = 21.9 Hz), 128.9, 128.7, 128.2, 127.9(d, J = 6.6 Hz), 127.4 (d, J = 7.1 Hz), 82.1 (d, J = 19.1 Hz), 73.7

(d, J = 3.1 Hz), 58.5, 45.1, 43.9 (d, J = 5.9 Hz), 34.3, 33.8, 26.2, 26.1, 25.8, 21.7, 21.3 (d, J = 2.2 Hz), 20.8; ³¹P NMR (162, MHz, CDCl₃) δ 117.3; HRMS (ES+) m/z calcd for C₃₁H₄₀O₂PS: 507.2487 [M + H]⁺, found 507.2506.

((1R,2S)-1-(tert-Butylthio)-1-mesityl-3-methoxypropan-2-yloxy)diphenylphosphine (7m). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 6m (26.7 mg, 0.09 mmol), DMAP (1.1 mg, 0.009 mmol), NEt₃ (15 µL, 0.11 mmol), and chlorodiphenylphosphine (17 μ L, 0.09 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography $(1 \times 3 \text{ cm}, 95\% \text{ hexane}, 5\% \text{ ethyl acetate})$ to yield **7m** as an oil (39 mg, 90% yield): $[\alpha]^{25}{}_{\mathrm{D}} = -23.4 (c \ 0.58, \text{CHCl}_3)$; IR (neat) 2920, 2894, 2861, 1479, 1470, 1457, 1435, 1126, 1093, 1068, 1054, 1028, 1000, 967, 947, 739, 697 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.39-7.44 (m, 2H), 7.25-7.31 (m, 3H), 7.09-7.13 (m, 1H), $6.96-7 (m, 2H), 6-6.64 (m, 4H), 4.6 (d, {}^{3}J = 11 Hz, 1H), 4.33-4$ (m, 1H), 3.8–3.86 (m, 2H), 3.14 (s, 3H), 2.37 (s, 3H), 2.36 (s, 3H), 2.20 (s, 3H), 1.26 (s, 9H); ^{13}C NMR (100 MHz, CDCl₃) δ 143.8, (d, J = 18.3 Hz), 142.2 (d, J = 13.2 Hz), 137.0, 136.2, 136.2,135.4, 131.0, 130.0 (d, J = 22 Hz), 128.9, 128.7, 128.2, 127.9 (d, J = 7.0 Hz), 127.4 (d, J = 6.6 Hz), 81.9 (d, J = 19.8 Hz), 73.3 (d, J = 2.85 Hz), 58.4, 44.6, 42.7 (d, J = 6.5 Hz), 31.3, 21.5 (d, J = 2.7 Hz), 20.8; ³¹P NMR (162, MHz, CDCl₃) δ 117.8; HRMS (ES+) m/z calcd for C₂₉H₃₈O₂PS 481.2330 [M + H]⁺, found 481.2314.

((1R,2S)-1-(Adamantvlthio)-1-mesityl-3-methoxypropan-2yloxy)diphenylphosphine (7n). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: **6n** (26 mg, 0.07 mmol), DMAP (0.8 mg, 0.007 mmol), NEt₃ (12 µL, 0.08 mmol), and chlorodiphenylphosphine (13 µL, 0.07 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1×3 cm, 95% hexane, 5% ethyl acetate) to yield **7n** as an oil (33 mg, 85% yield): $[\alpha]^{26}_{D}$ -45.7 (c 0.42, CHCl₃); IR (neat) 2903, 2848, 1479, 1450, 1434, 1376, 1343, 1300, 1254, 1126, 1093, 1053, 946, 909, 888, 737, 697 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.39–7.44 (m, 1H), 7.23-7.30 (m, 3H), 7.09-7.12 (m, 1H), 6.96-7 (m, 2H), 6.59-6.63 (m, 4H), 4.62 (d, ${}^{3}J = 11.1$ Hz, 1H), 4.33-4.40 (m, 1H), 3.86 (dxd, ${}^{2}J = 10$ Hz, ${}^{3}J = 2.1$ Hz, 1H), 3.80 (dxd, $^{2}J = 10$ Hz, $^{3}J = 4.4$ Hz, 1H), 3.14 (s, 3H), 2.38 (s, 3H), 2.37 (s, 3H), 2.2 (s, 3H), 1.98 (br s, 3H), 1.59–1.87 (m, 12H); ¹³C NMR (100 MHz, CDCl₃) δ 143.9 (d, J = 18 Hz), 142.3 (d, J =13.9 Hz), 137.0, 136.6, 135.8, 135.3, 131.9, 130.0 (d, J = 22 Hz), 130.0 (d, J = 22 Hz), 128.9, 128.6, 128.1, 127.8 (d, J = 7 Hz), 127.4 (d, J = 6.6 Hz), 82.0 (d, J = 19.6 Hz), 73.5 (d, J = 2.7 Hz),58.4, 43.7, 40.0 (d, J = 6.4 Hz), 36.2, 29.7, 21.8, 21.5 (d, J =2.2 Hz), 20.8; ³¹P NMR (162, MHz, CDCl₃) δ 117.8; HRMS (ES+) m/z calcd for C₃₅H₄₃O₂NaPS 581.2619 [M + Na]⁺, found 581.2620.

((1R,2S)-1-(tert-Butylthio)-1-mesityl-3-(trityloxy)propan-2yloxy)diphenylphosphine (10b). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 9b (79 mg, 0.15 mmol), DMAP (1.9 mg, 0.015 mmol), NEt₃ (25 µL, 0.18 mmol), and chlorodiphenylphosphine (28 μ L, 0.15 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography $(1 \times 3 \text{ cm}, 95\% \text{ hexane},$ 5% ethyl acetate) to yield 10b as an oil (98 mg, 92% yield): [α]²⁶_D+13.1 (c 0.33, CHCl₃); IR (neat) 3056, 2958, 2917, 1738, 1596, 1488, 1447, 1363, 1261, 1219, 1157, 1083, 1048, 1022, 998, 966, 943, 913, 807, 740, 694 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) & 7.03-7.30 (m, 21H), 6.9-6.94 (m, 2H), 6.72-6.75 (m, 2H), 6.45 (s, 1H), 6.38 (s, 1H), 4.4-4.47 (m, 1H), 4.21 (d, ${}^{3}J = 11$ Hz, 1H), 3.99 (dxd, ${}^{2}J = 10$ Hz, ${}^{3}J = 1.5$ Hz, 1H), 3.22 $(dxd, {}^{2}J = 10 \text{ Hz}, {}^{3}J = 6.5 \text{ Hz}, 1\text{H}), 2.24 (s, 3\text{H}), 2.22 (s, 3\text{H}),$ 2.05 (s, 3H), 0.94 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 144.0, 143.6 (d, J = 16 Hz), 142.4 (d, J = 16.8 Hz), 137.1, 135.7, 135.5, 135.4, 131.0, 130.4 (, d, J = 24 Hz), 129.7 (d, J = 21.67 Hz), 128.9, 128.9, 128.3, 127.8 (d, J = 6.6 Hz), 127.5, 127.3 (d, J = 7.4 Hz), 126.7, 87.2, 82.9 (d, J = 20.4 Hz), 67.2 (d, J = 2.32 Hz), 44.1, 43.6 (d, J = 5.2 Hz), 31.2, 21.7 (d, J = 3.1 Hz), 20.7; ³¹P NMR (162, MHz, CDCl₃) δ 116.8; HRMS (ES+) m/z calcd for C₄₇H₅₀O₂PS 709.3269 [M + Na]⁺, found 709.3268.

((1R,2R)-1-(Adamantylthio)-1-phenyl-3-(trityloxy)propan-2yloxy)diphenylphosphine (10a-syn). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 9a-syn (100 mg, 0.18 mmol), DMAP (2.2 mg, 0.018 mmol), NEt₃ (30 µL, 0.21 mmol), and chlorodiphenylphosphine (33 μ L, 0.18 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1×3 cm, 95% hexane, 5% ethyl acetate) to yield **10a-syn** as a white solid (120 mg, 90% yield): $[\alpha]^{27}_{D}$ -73.1 (c 0.51, CHCl₃); IR (neat) 3054, 2901, 1597, 1490, 1447, 1434, 1342, 1071, 741, 694 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.37-7.41 (m, 4H), 7.32-7.35 (m, 6H), 7.27-7.29 (m, 3H), 7.18-7.24 (m, 14H), 7.08-7.1 (m, 3H), 4.22-4.28 (m, 1H), 3.36 $(dxd, {}^{2}J = 9.5 \text{ Hz}, {}^{3}J = 5.4 \text{ Hz}, 1\text{H}), 3.16 (dxd, {}^{2}J = 10.4 \text{ Hz},$ ${}^{3}J = 5.4$ Hz, 1H), 1.88 (br s, 3H), 1.51–1.69 (br m, 12H); ${}^{13}C$ NMR (100 MHz, CDCl₃) δ 143.9, 142.8 (d, J = 19.2 Hz), 142.9, 142.2 (d, J = 15.9 Hz), 131.0 (d, J = 22.7 Hz), 130.5 (d, J = 22.2 Hz), 129.0, 128.9, 128.8, 128.7, 128.0 (d, J = 7.3 Hz), 128.0 (d, J = 7.3 Hz), 127.8, 127.7, 126.8, 126.5, 86.1, 84.6 (d, J =18.3 Hz), 65.3 (d, J = 4.3 Hz), 47.6 (d, J = 5.2 Hz), 45.7, 43.7, 36.2, 29.7; ³¹P NMR (162 MHz, CDCl₃) δ 117.2; HRMS (ES+) m/z calcd for C₅₀H₄₉O₂PSNa 767.3089 [M], found 767.3052.

((1R,2R)-1-(tert-Butylthio)-1-phenyl-3-(trityloxy)propan-2yloxy)diphenylphosphine (10b-syn). The following reagents were combined in the amounts indicated according to the general procedure for the phosphinite incorporation: 9b-syn (100 mg, 0.19 mmol), DMAP (2.4 mg, 0.019 mmol), NEt₃ (32 µL, 0.23 mmol), and chlorodiphenylphosphine (35 µL, 0.19 mmol). The resulting slurry was loaded onto a plug of silica and purified by flash chromatography (1×3 cm, 95% hexane, 5% ethyl acetate) to yield **10b**-syn as a white solid (112 mg, 83% yield): $\left[\alpha\right]^{2/2}$ - 52.5 (c 0.51, CHCl₃); IR (neat) 3057, 2959, 2914, 2860, 1597, 1448, 1260, 1064, 984, 926, 799, 739, 693 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.53–7.63 (m, 4H), 7.07–7.32 (m, 21H), 6.62 (s, 1H), 6.57 (s, 1H), 4.54–4.62 (m, 1H), 4.24 (d, ${}^{3}J = 8.9$ Hz, 1H), 2.96-3.05 (m, 2H), 2.33 (s, 3H), 2.17 (s, 3H), 2.15 (s, 3H), 0.95 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 143.7, 143.1 (d, J = 17.6 Hz), 142.7 (d, J = 16.1 Hz), 137.2, 135.8, 135.5, 134.2, 132.1 (d, J = 22.7 Hz), 131.4, 130.4 (d, J = 22.3 Hz), 129.0, 128.9,128.6, 127.9 (d, J = 6.9 Hz), 127.7 (d, J = 7.3 Hz), 127.4, 126.6, 87.1, 84.0 (d, J = 19.7 Hz), 66.4 (d, J = 2.9 Hz), 46.8 (d, J =4.4 Hz), 43.83, 31.1, 22.2, 21.5, 20.7; ³¹P NMR (162, MHz, CDCl₃) δ 118.6; HRMS (ES+) m/z calcd for C₄₇H₅₀O₂PS 709.3269 [M - H]⁻, found 709.3250.

((1R,2R)-1-(Adamantylthio)-1-phenyl-3-(trityloxy)propan-2yloxy)diphenylphosphinepalladium(II) Chloride (10a-syn) · PdCl₂. To a Schlenk flask containing the thiophosphinite ligand 10a-syn (34.3 mg, 0.046 mmol) in CH₂Cl₂ (1 mL) was added (MeCN)₂PdCl₂ (11.9 mg 0.046 mmol), and the reaction was stirred for 40 min. The solution was concentrated in vacuo, and the residue was diluted with CH₂Cl₂ (0.6) mL. Then Et₂O (10 mL) was added rapidly with stirring to precipitate PdCl₂-10a-syn, which was filtered and dried in vacuo overnight (42 mg, 99% yield): mp >200 °C; $[\alpha]^{2/}_{D}$ -108.2 (c 0.54, CH₂Cl₂); IR (neat) 3054, 2901, 1597, 1490, 1447, 1434, 1342, 1071, 741, 694 cm⁻¹; ¹H NMR (400 MHz, CD₂Cl₂) δ 8.28-8.33 (m, 2H), 7.81-7.86 (m, 2H), 7.60-7.65 (m, 1H), 7.52-7. 57 (m, 2H), 7.40–7.43 (m, 1H), 7.08–7.33 (m, 22H), 4.41 (d, $J^3 = 10.6$ Hz, 1H), 3.82–3.88 (m, 1H), 2.85 (dxd, $^2J = 10.6$ Hz, $^3J = 2.2$ Hz, 1H), 2.39 (dxd, ${}^{2}J = 10.6$ Hz, ${}^{3}J = 2.2$ Hz, 1H), 2.22 (br d, 3H), 1.99 (br s, 6H), 1.55–1.64 (br m, 6H); ¹³C NMR (100 MHz,

CDCl₃) δ 147.3 (d, J = 26.6 Hz), 147.3 (d, J = 27 Hz), 143.4, 136.7, 134.5 (d, J = 13 Hz), 133.1, 133.2, 129.9, 129.6, 129.2 (d, J = 12.2 Hz), 128.8 (d, J = 17.8 Hz), 128.7, 128.6, 128.5, 128.2, 127.6, 87.1, 80.6, 62.9 (d, J = 6.8 Hz), 60.6, 47.6 (d, J = 5.2 Hz), 44.3, 42.7, 35.8, 30.8; ³¹P NMR (162 MHz, CD₂Cl₂) δ 110.8. X-ray quality crystals were grown by slow vapor diffusion of Et₂O into a solution of **PdCl₂-10a**-*syn* in CH₂Cl₂ to yield orthorhombic crystals. A suitable crystal (0.15 × 0.15 × 0.06 mm) was chosen for X-ray diffraction analysis; see the Supporting Information.

General Procedure for the Palladium-Catalyzed Allylic Alkylation Reaction. To a Schlenck flask containing ligand 10b-syn (4.215 mg, 0.006 mmol) and [(C₃H₅)PdCl]₂ (1.088 mg, 0.006 mmol Pd) was added CH₂Cl₂ (0.6 mL). After 1 h of stirring at rt, 1,3-diphenylpropenyl acetate (60 mg, 0.238 mmol) dissolved in CH₂Cl₂ (0.6 mL) was added, followed by dimethyl malonate (82 µL, 0.71 mmol), N,O-bis(trimethylsilyl)acetamide (BSA) (177 μ L, 0.71 mmol), and a pinch of KOAc. The mixture was stirred at room temperature for 20 min and then was diluted with diethyl ether and washed with saturated NH₄Cl_(aq). The organic layers were dried over MgSO4 and filtered, and the solvents were evaporated in vacuo. The residue was purified by flash chromatrography (85% hexane, 15% EtOAc) to afford (S,E)-dimethyl 2-(1,3-diphenylallyl)malonate as a colorless oil. The absolute configuration of the product was assigned by comparing the sign of its specific rotation with literature data.⁴ HPLC analysis (Chiralcel-AD-H, hexane/2-propanol 95:5, 1 mL/min, 254 nm, $t_R = 21$ min, $t_S = 24$ min) showed the ee to be 96%.

(*S*)-Methyl 2-Carbomethoxy-3,5,5-triphenylpent-4-enoate (P2). The following reagents were combined in the amounts indicated according to the general procedure for the allylic alkylation with dimethyl malonate: 9a-syn (2.48 mg, 0.003 mmol), $[(C_3H_5)PdCl]_2$ (0.54 mg, 0.003 mmol), 1,3,3-triphenylpropenyl acetate (S2) (39 mg, 0.12 mmol), dimethyl malonate (41 μ L, 0.36 mmol), BSA (88 μ L, 0.36 mmol), and a pinch of potassium acetate. The product was purified by flash chromatography (9:1 hexane/ethyl acetate) to yield P3 as an oil (26 mg, 54% yield). The absolute configuration of the product was assigned by comparing the sign of its specific rotation with literature data. [α]²⁵_D = -182 (*c* 0.3, CH₂Cl₂); [α]²⁵_D = -186 (*c* 0.44, CH₂Cl₂) for optically pure P2. HPLC analysis (AD-H column, 0.3 mL/min *n*-hexane/2-propanol 97:3: t_R = 44.7 min, t_S = 46.7 min) showed the ee to be 97%.

(S)-Methyl 2-Carbomethoxy-3-methyl-5,5-diphenylpent-4enoate (P3). The following reagents were combined in the amounts indicated according to the general procedure for the allylic alkylation with dimethyl malonate: 9a (5.3 mg, 0.007 mmol), [(C₃H₅)PdCl]₂ (1.08 mg, 0.006 mmol), 4,4-diphenylbutenyl-2-acetate (S3) (63 mg, 0.24 mmol), dimethyl malonate (82 μ L, 0.71 mmol), BSA (177 μ L, 0.71 mmol), and a pinch of potassium acetate. The product was purified by flash chromatography (9:1 hexane/ethyl acetate) to yield P3 as an oil (33 mg, 40% yield). HPLC analysis (OD-H column, 1 mL/min *n*-hexane/2-propanol 99:1: $t_R = 9.1 \text{ min}, t_S = 10.6 \text{ min}$) showed the ee to be 82%.

General Procedure for the Palladium-Catalyzed Allylic Amination Reaction. To a Schlenk flask containing ligand 10b-syn (4.215 mg, 0.006 mmol) and $[(C_3H_5)PdCl]_2$ (0.725 mg, 0.004 mmol Pd) was added CH₂Cl₂ (0.5 mL). After 1 h of stirring at rt, 1,3-diphenylpropenyl acetate (40 mg, 0.16 mmol) dissolved in CH₂Cl₂ (0.5 mL) was added, followed by benzylamine (52 μ L, 0.476 mmol). The mixture was stirred at room temperature for 4 h. Then the mixture was diluted with diethyl ether and washed with saturated NH₄Cl_{aq}. The organic layers were dried over MgSO₄ and filtered and the solvents were evaporated *in vacuo*. The residue was purified by flash chromatrography (87% hexane, 13% EtOAc) to afford (*R*,*E*)-*N*-benzyl-1,3-diphenyl-prop-2-en-1-amine (97% yield) as a colorless oil. The absolute configuration of the product was assigned by comparing the sign

of its specific rotation with literature data.^{5j} HPLC analysis (Chiralcel-OJ, hexane/2-propanol 87:13, 0.5 mL/min, 254 nm, $t_s = 14.2 \text{ min}, t_R = 17.5 \text{ min}$) showed the ee to be 94%.

General Procedure for the Palladium-Catalyzed Allylic Etherification Reaction. To a Schlenk flask containing ligand 10b-sym (3.717 mg, 0.005 mmol) and $[(C_3H_5)PdCl]_2$ (0.914 mg, 0.005 mmol Pd) was added toluene (0.6 mL). After 1 h of stirring at rt, 1,3-diphenylpropenyl acetate (31.5 mg, 0.125 mmol) dissolved in CH₂Cl₂ (0.5 mL) was added, followed by cesium carbonate (122 mg, 0.375 mmol) and benzyl alcohol (39 μ L, 0.375 mmol). The mixture was stirred at 0 °C for 3 h, and then the mixture was diluted with ethyl acetate and washed with saturated NH₄Cl_{aq}. The organic layers were dried over MgSO₄ and filtered, and the solvents were evaporated under vacuum. The residue was purified by flash chromatography (95% hexane, 5% Et₂O) to afford (*R,E*)-(3-(benzyloxy)prop-1-ene-1,3-diyl)dibenzene (96% yield) as a colorless oil. The absolute configuration of the product was assigned by comparing the sign of its specific rotation with literature data.^{5t} HPLC analysis (Chiralcel-OJ-H, hexane/2-propanol 98:2, 0.75 mL/min, 254 nm, $t_S = 27.9 \text{ min}$, $t_R = 34 \text{ min}$) showed the ee to be 94%.

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Supporting Information Available: Copies of ¹H and ¹³C NMR spectra. Crystallographic data for (10a-syn)·PdCl₂ and $4 \text{ m} \cdot \text{PdCl}_2$ (CIF). Cartesian coordinates and energies of TS C2 and TS C3 calculated at the BLYP/6-31G (d) level. This material is available free of charge via the Internet at http://pubs.acs.org.